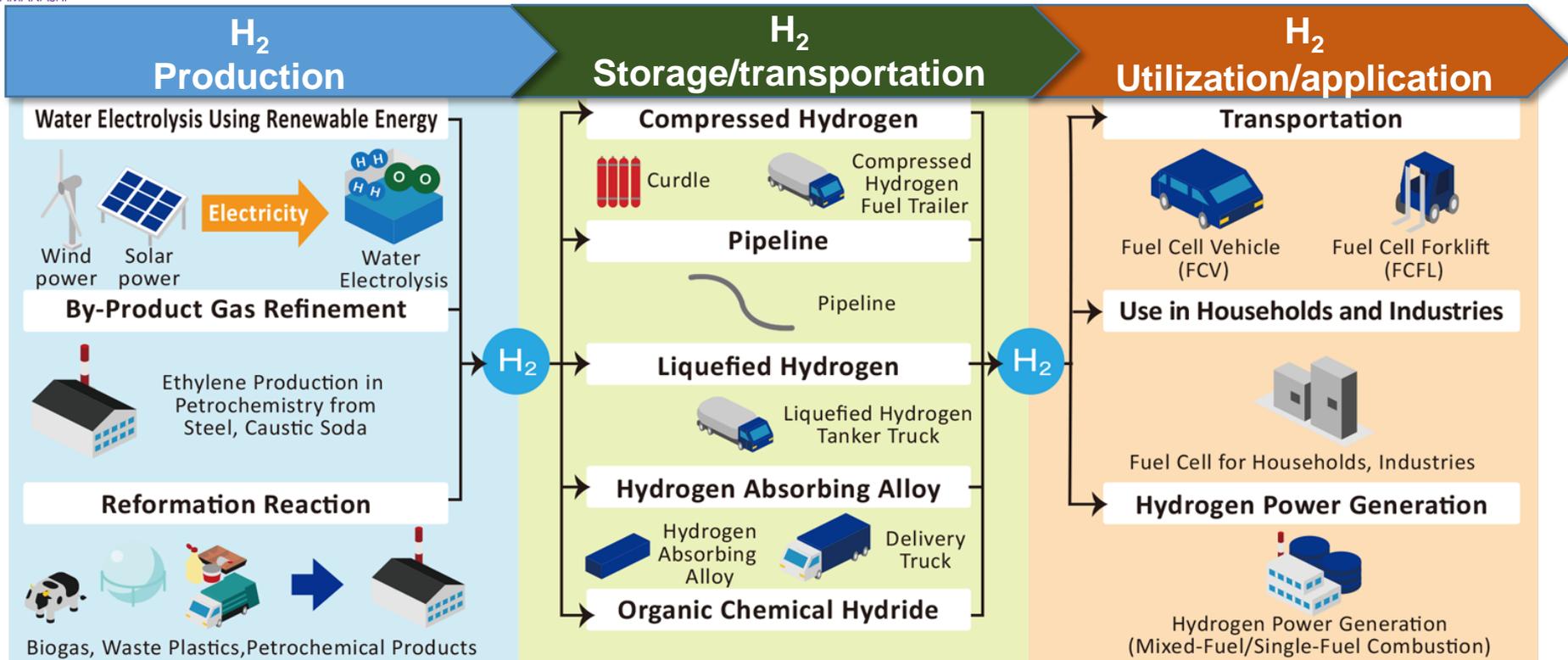


# Highly Durable and Active Cathode Catalysts using Niobium for Polymer Electrolyte Fuel Cells

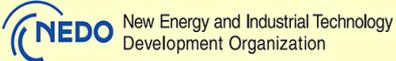
**Katsuyoshi Kakinuma**  
**Fuel Cell Nanomaterials Center,  
University of Yamanashi**





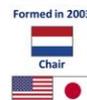
[https://www.env.go.jp/seisaku/list/ondanka\\_saisei/lowcarbon-h2-sc/en/demonstration-business/index.html](https://www.env.go.jp/seisaku/list/ondanka_saisei/lowcarbon-h2-sc/en/demonstration-business/index.html)

## Hydrogen Energy Ministerial Meeting



[https://hem-2021.nedo.go.jp/\\_en/](https://hem-2021.nedo.go.jp/_en/)

## International Partnership for Hydrogen and Fuel Cells in the Economy



[www.iphe.net](http://www.iphe.net)

Working Groups: Education & Outreach  
Regulations, Codes, Standards & Safety



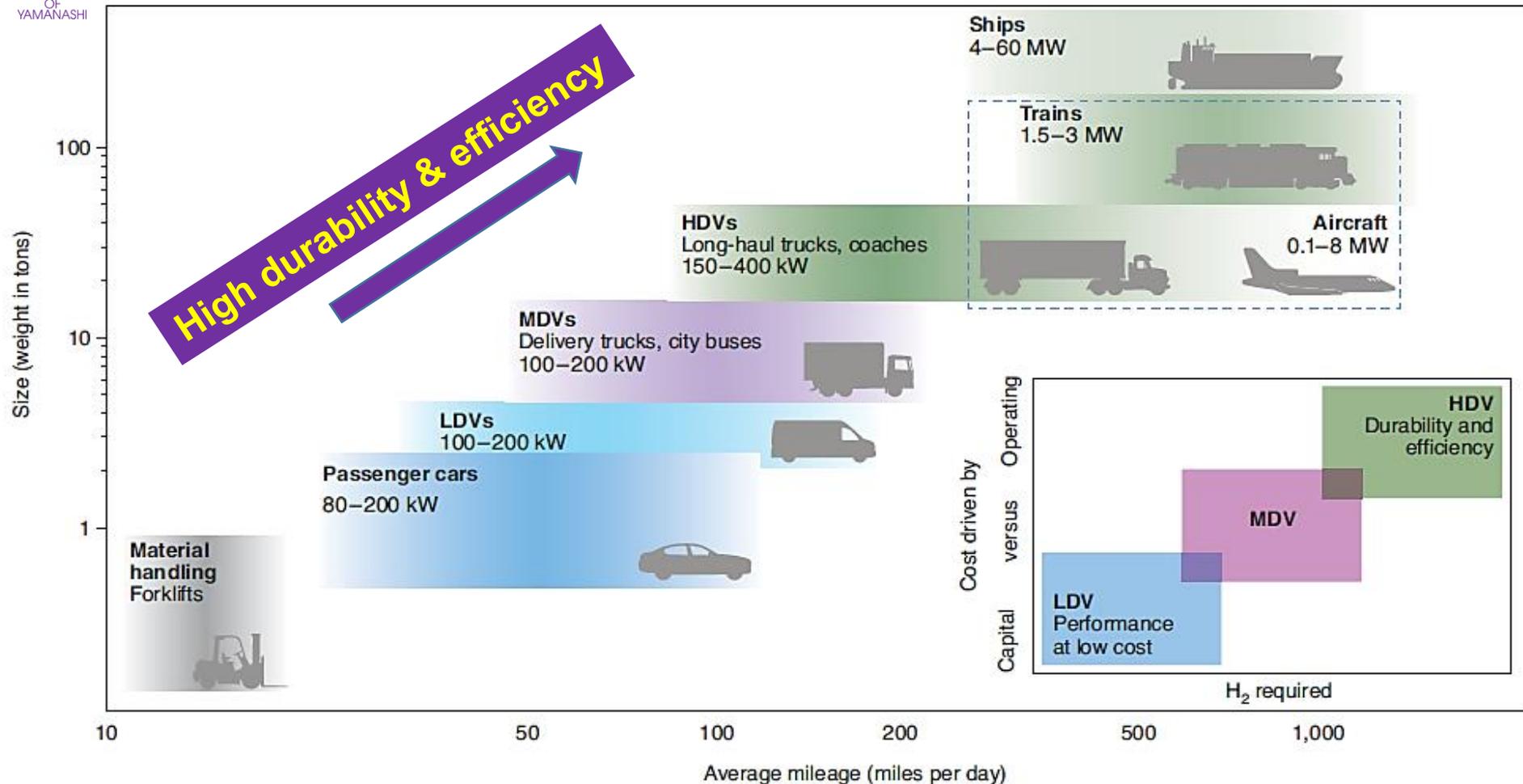
Task Forces:

1. H<sub>2</sub> Production Analysis
2. H<sub>2</sub> Trade Rules

And by interacting with International Initiatives and Organizations including the HEM, IEA (ETPD, TCPs), CEM/MI, WEF, HC & Others



<https://www.iphe.net>



D.A. Cullen, K.C. Neyerlin, R.K. Ahluwalia, R. Mukundan, K.L. More, R.L. Borup, A.Z. Weber, D.J. Myers, A. Kusoglu, Nature Energy 6 (2021) 462-474. Copyright permission from Springer Nature

PEFCs will be developed to meet the requirements of the heavy-duty vehicle market (in transportation) with the higher efficiency and durability.

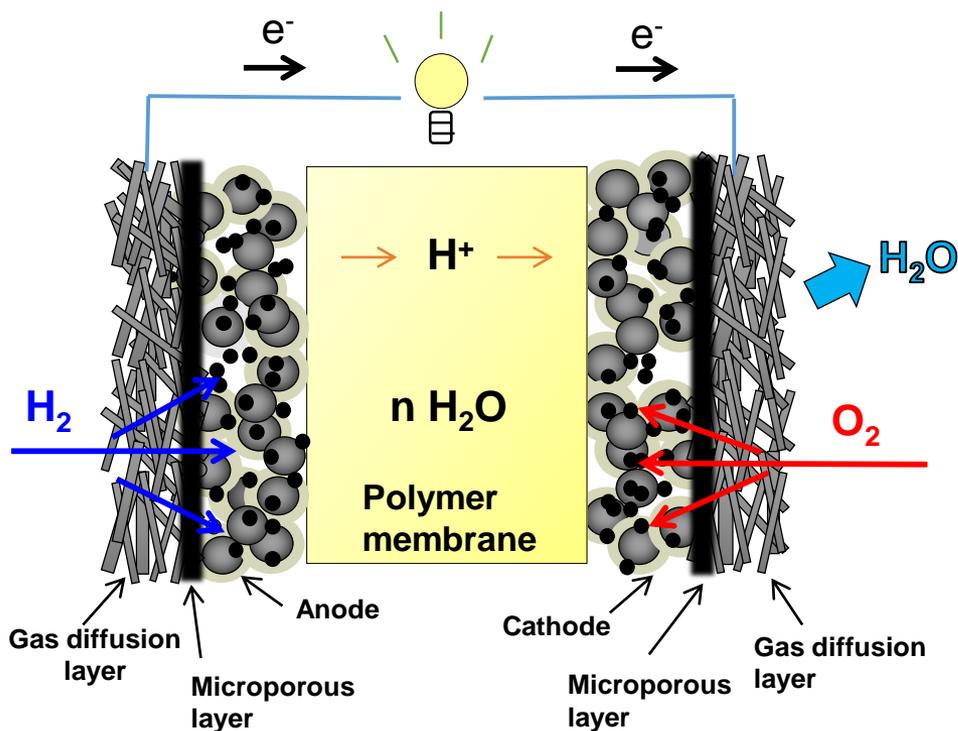
Introduction

Design concept

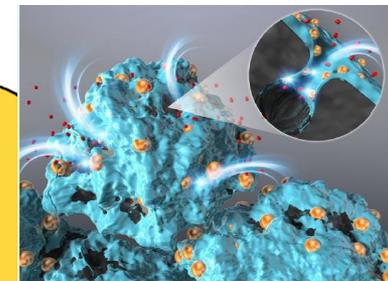
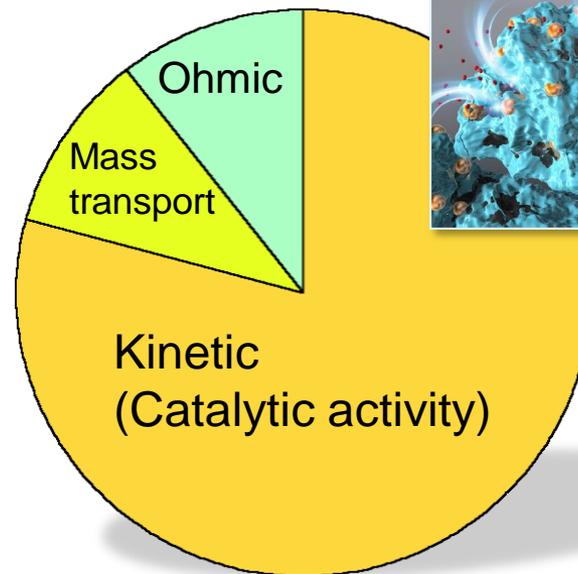
Performance

Future plan

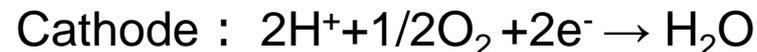
A. Kobayashi et al.  
ACS Appl. Energy Mater. 4 (2021) 2307.



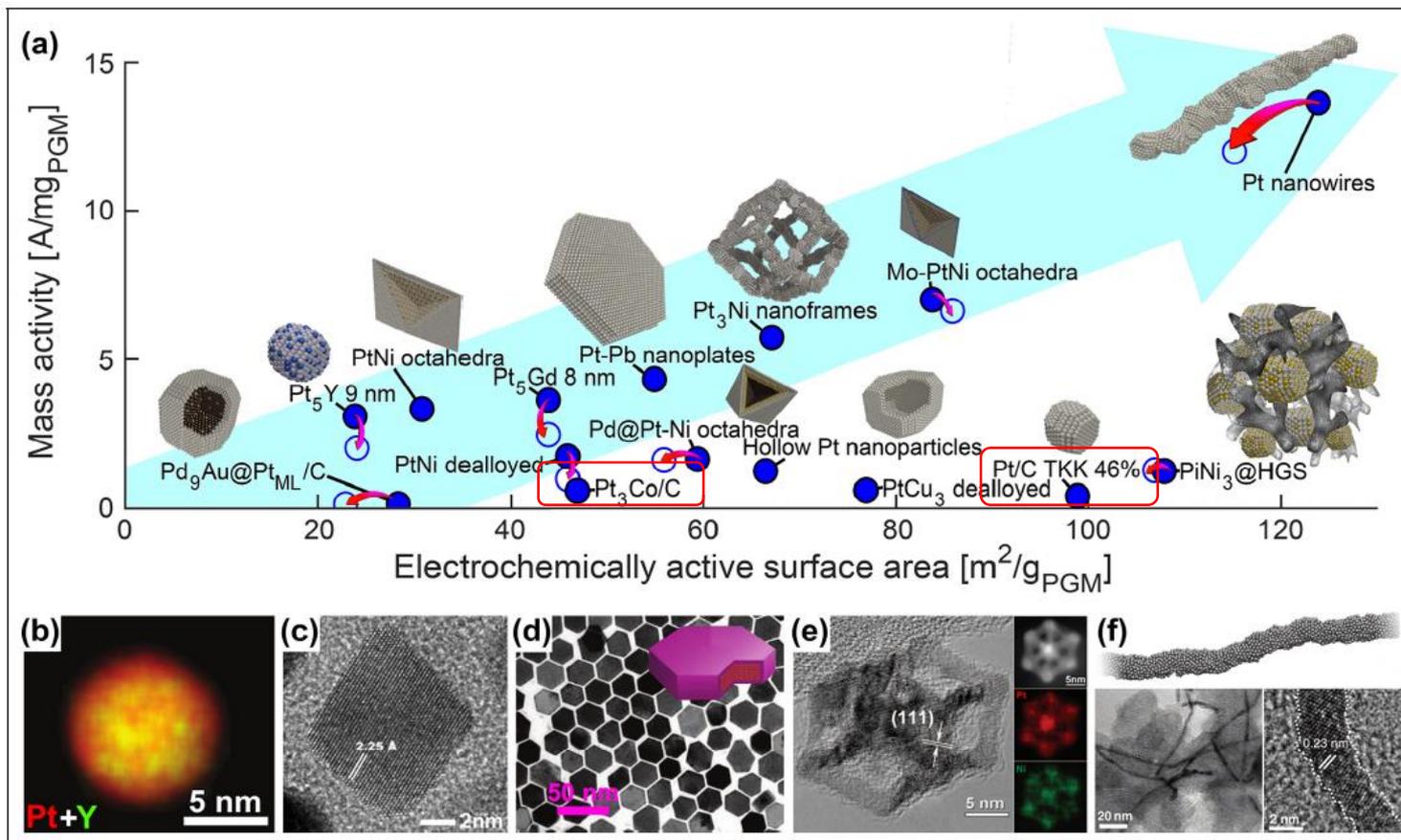
Polymer electrolyte fuel cell



Overpotential (@ 1.0 A cm<sup>-2</sup> 80°C)

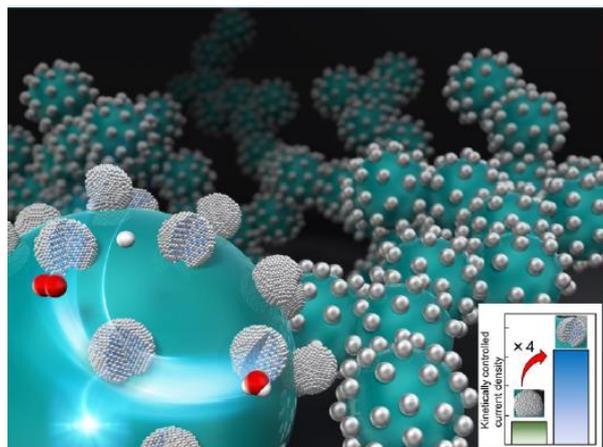


The new electrocatalysts with higher activity and durability are required.

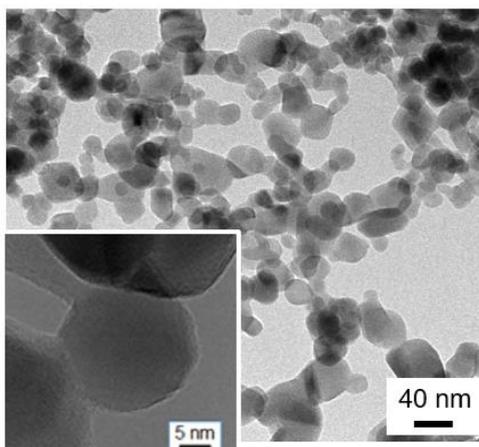


M. Escudero-Escribano, K.D. Jensen, A.W. Jensen Current Opinion Electrochem., 8 (2018) 135-146.  
Copyright permission from Elsevier

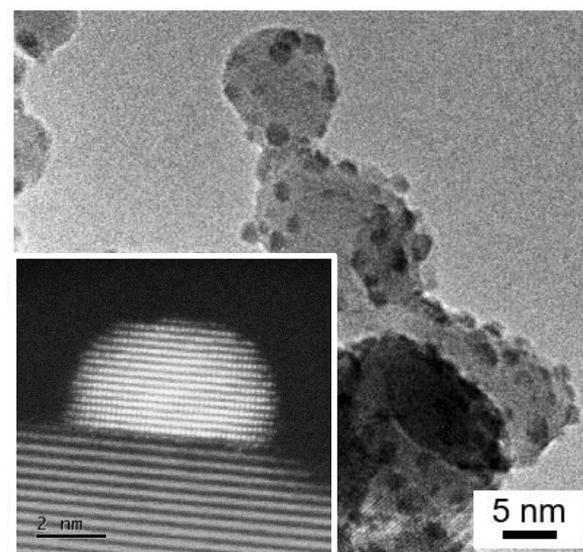
Pt alloys, nanorods and nanowires are one of the candidate electrocatalysts for fuel cells.



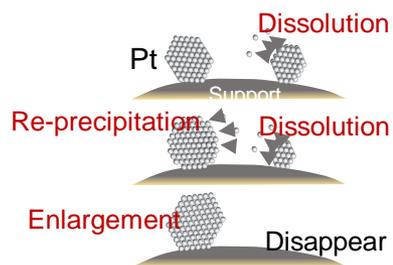
ACS Appl. Energy Mater. 2020, 3, 6922



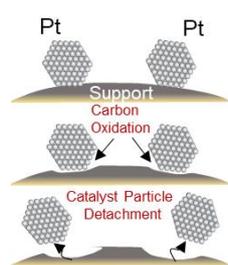
Electrochim. Acta, 2011, 56, 2881.



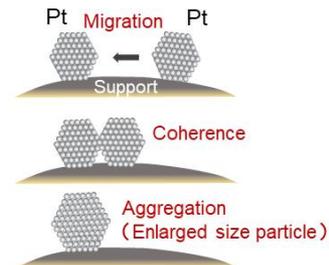
## Dissolution and enlargement



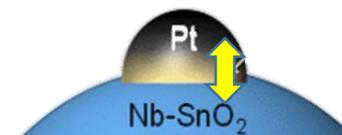
## Catalyst particle detachment



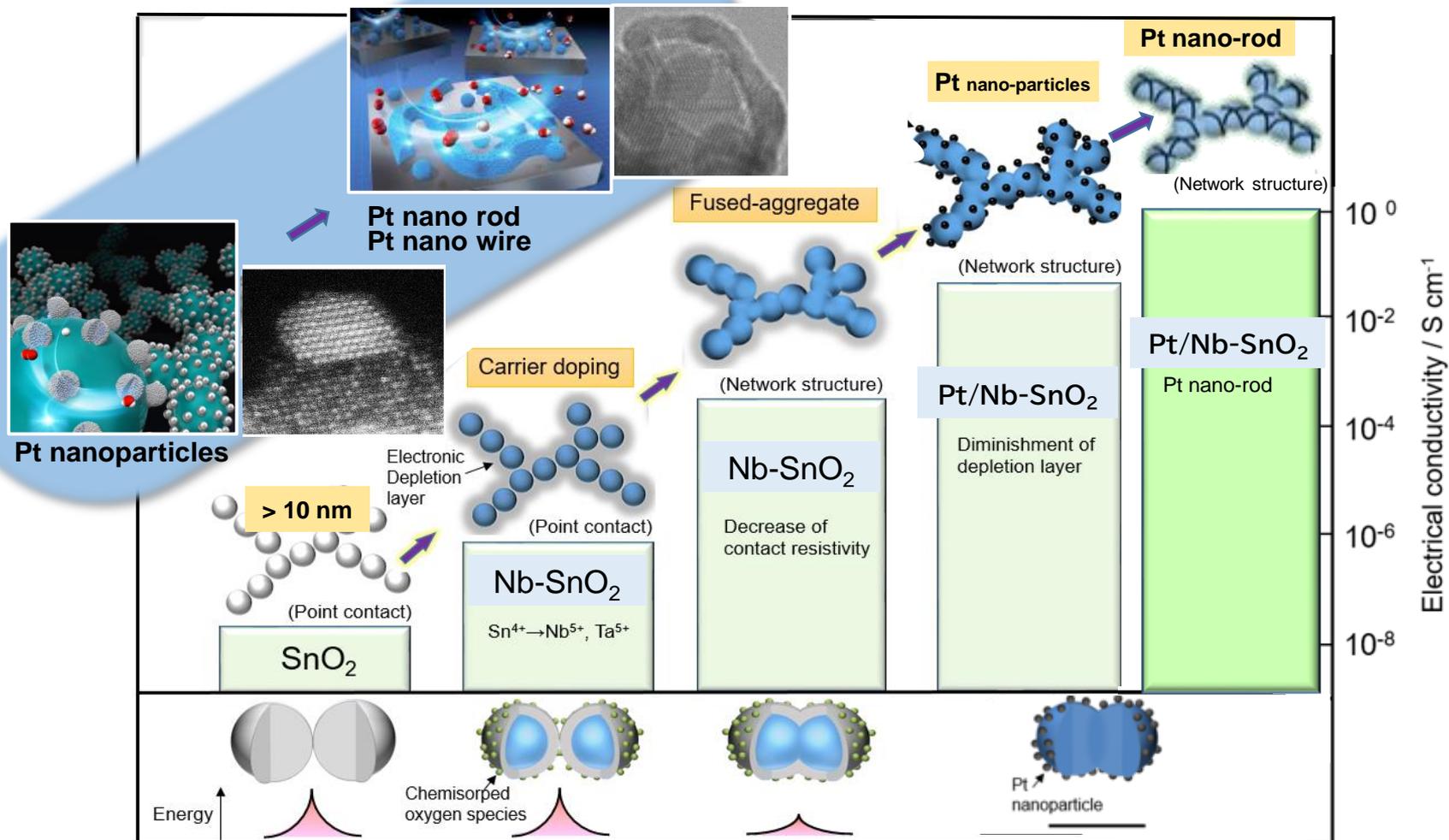
## Aggregation



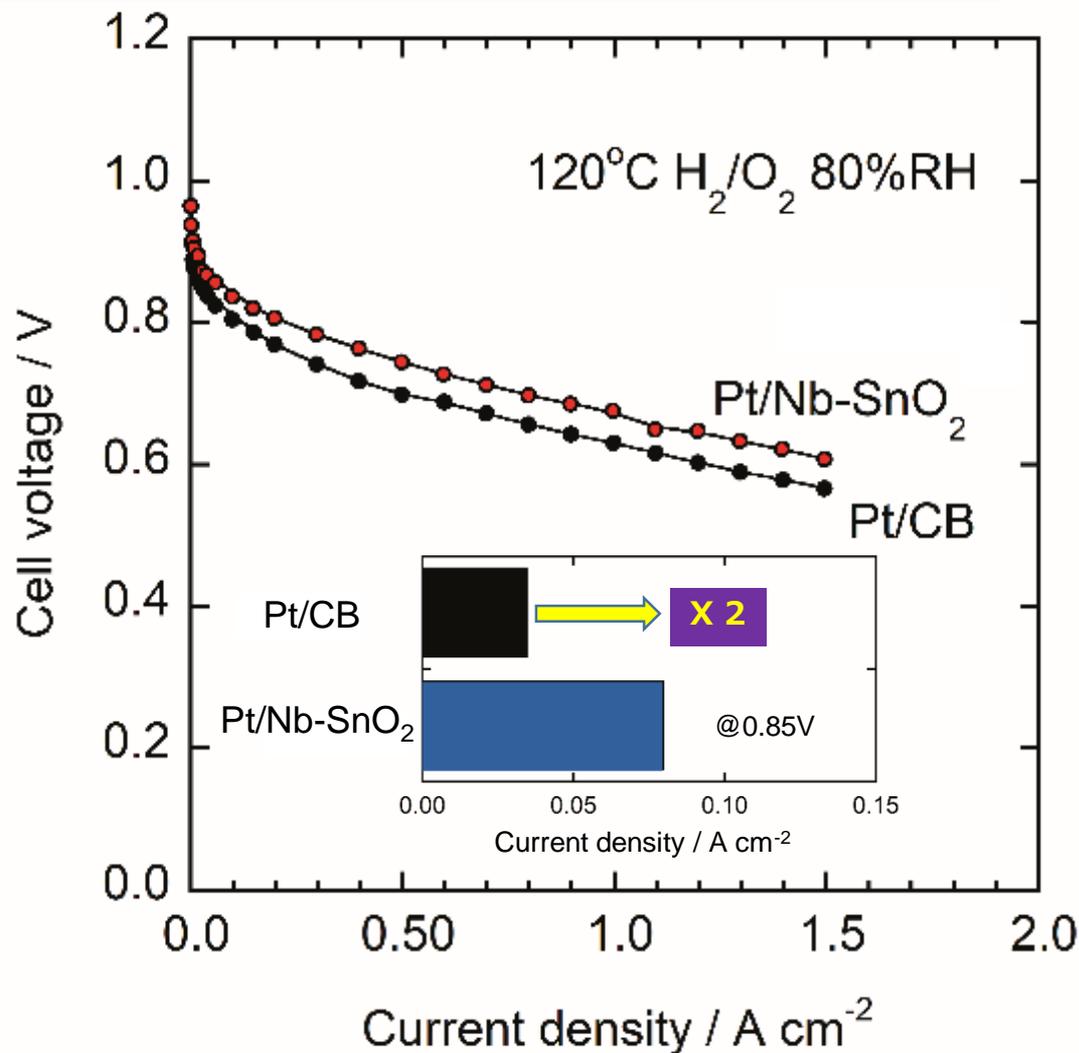
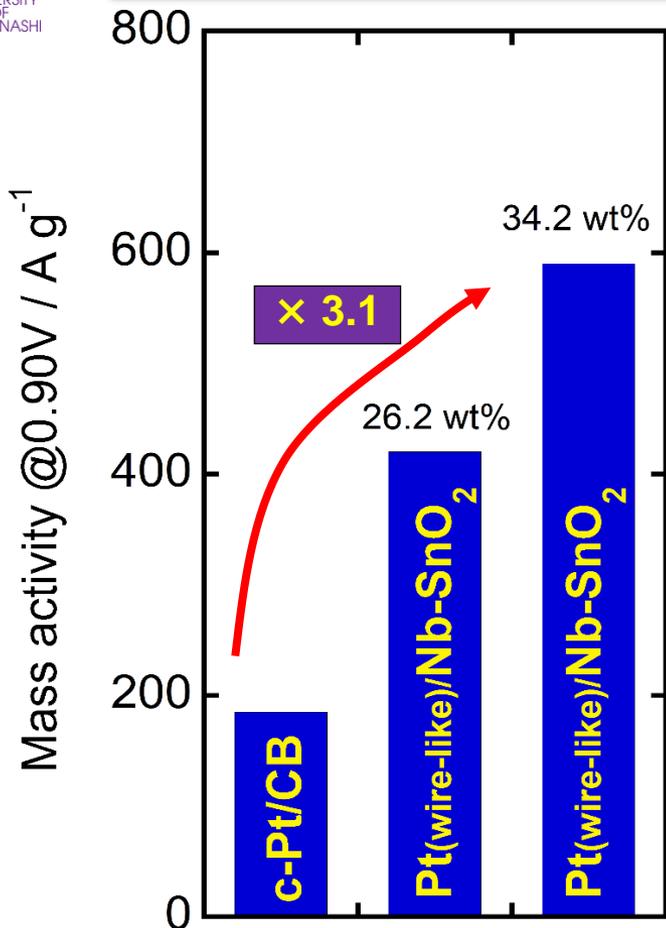
## Metal - support interaction



Startup/shutdown & load cycle durability of Pt/Nb-SnO<sub>2</sub> is superior to that of Pt/GCB, which relies on the strong bonding between Pt and Nb-SnO<sub>2</sub> and well size control of Pt particle



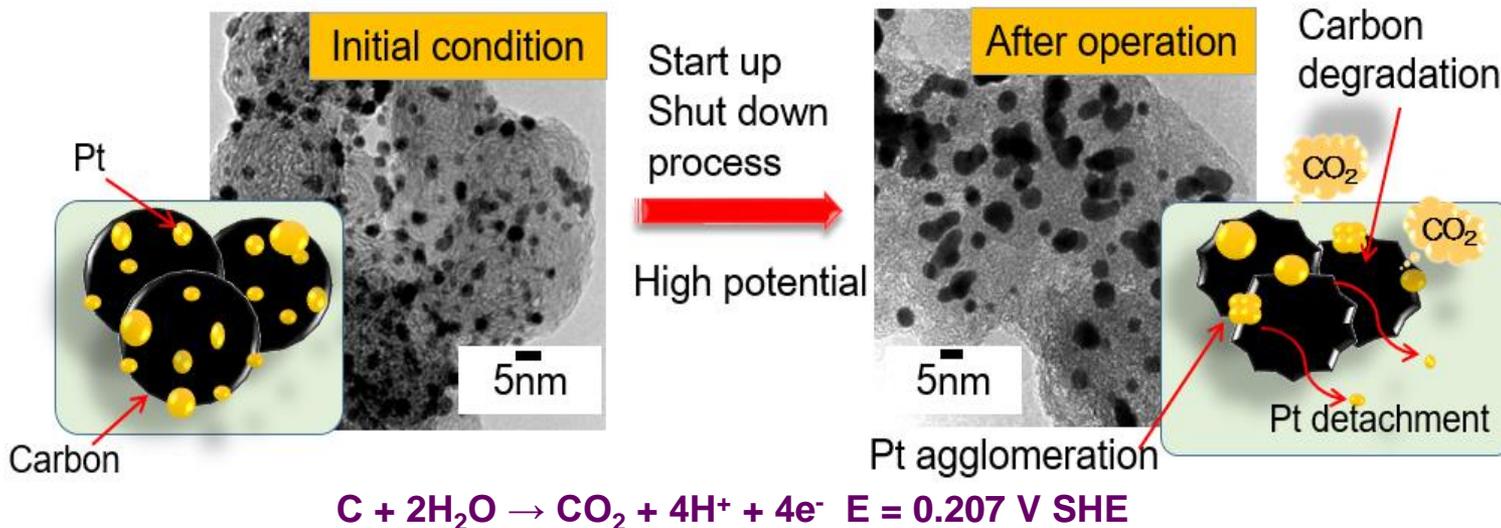
K.Kakinuma, et al. ACS Appl. Mater & Interfaces 11 (2019) 34957  
 G. Shi, et al. ACS Catal.11 (2021) 5222.



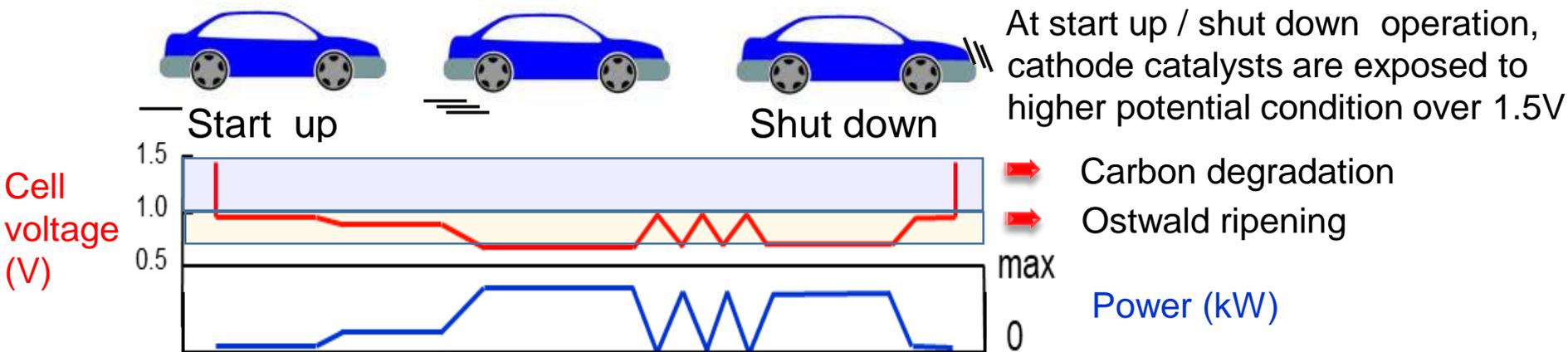
G. Shi, et al. ACS Catal.11 (2021) 5222.

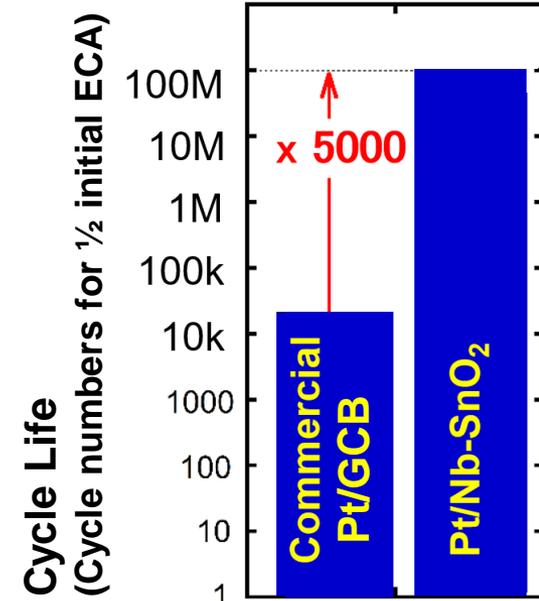
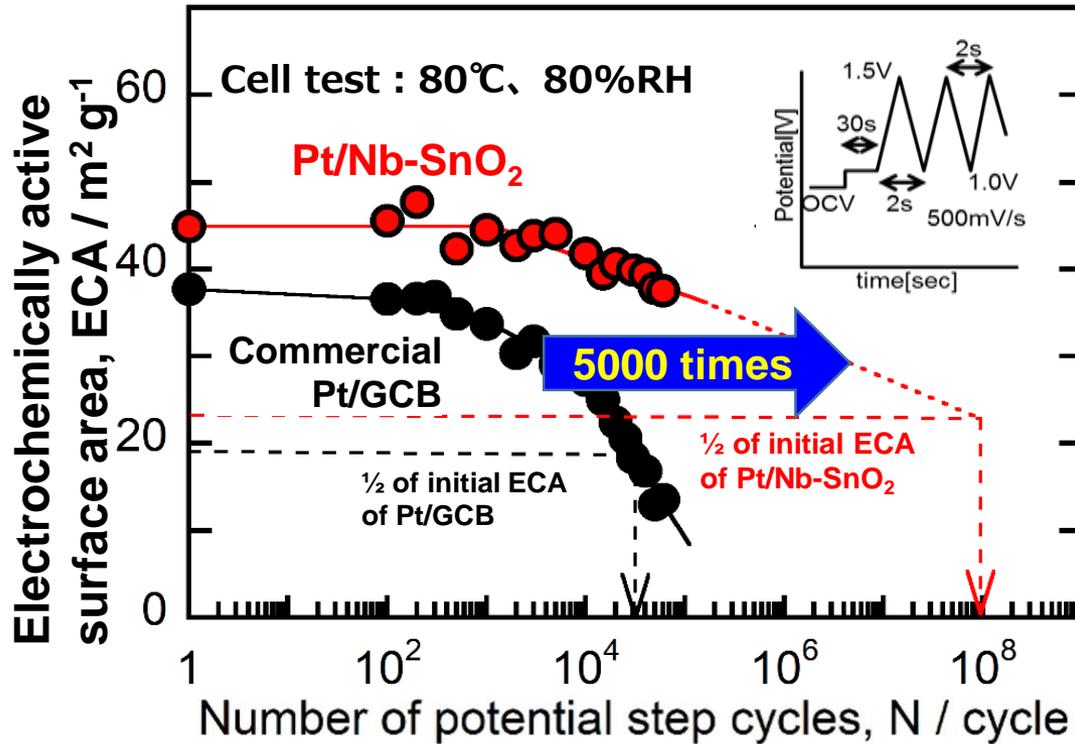
Single cell performances (power/durability) using Pt/Nb-SnO<sub>2</sub> catalyst layers is superior to those using current Pt supported on carbon (Pt/CB).

## Carbon support : Intrinsic thermodynamic instability



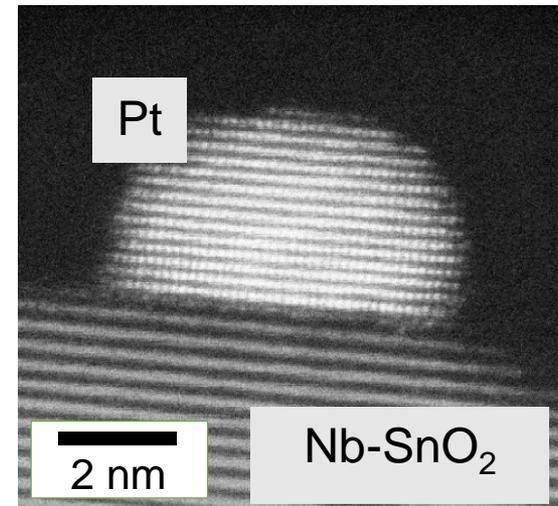
## FCV operation

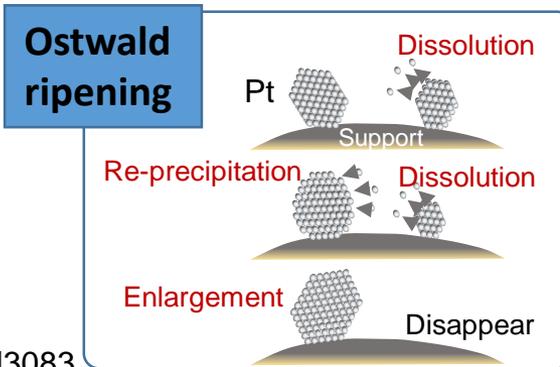
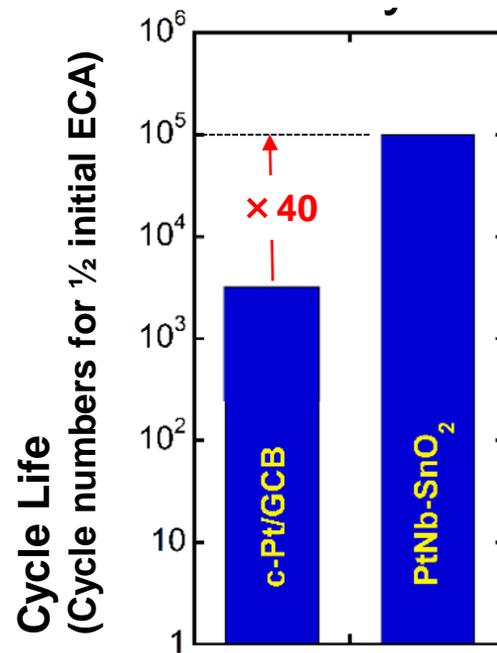
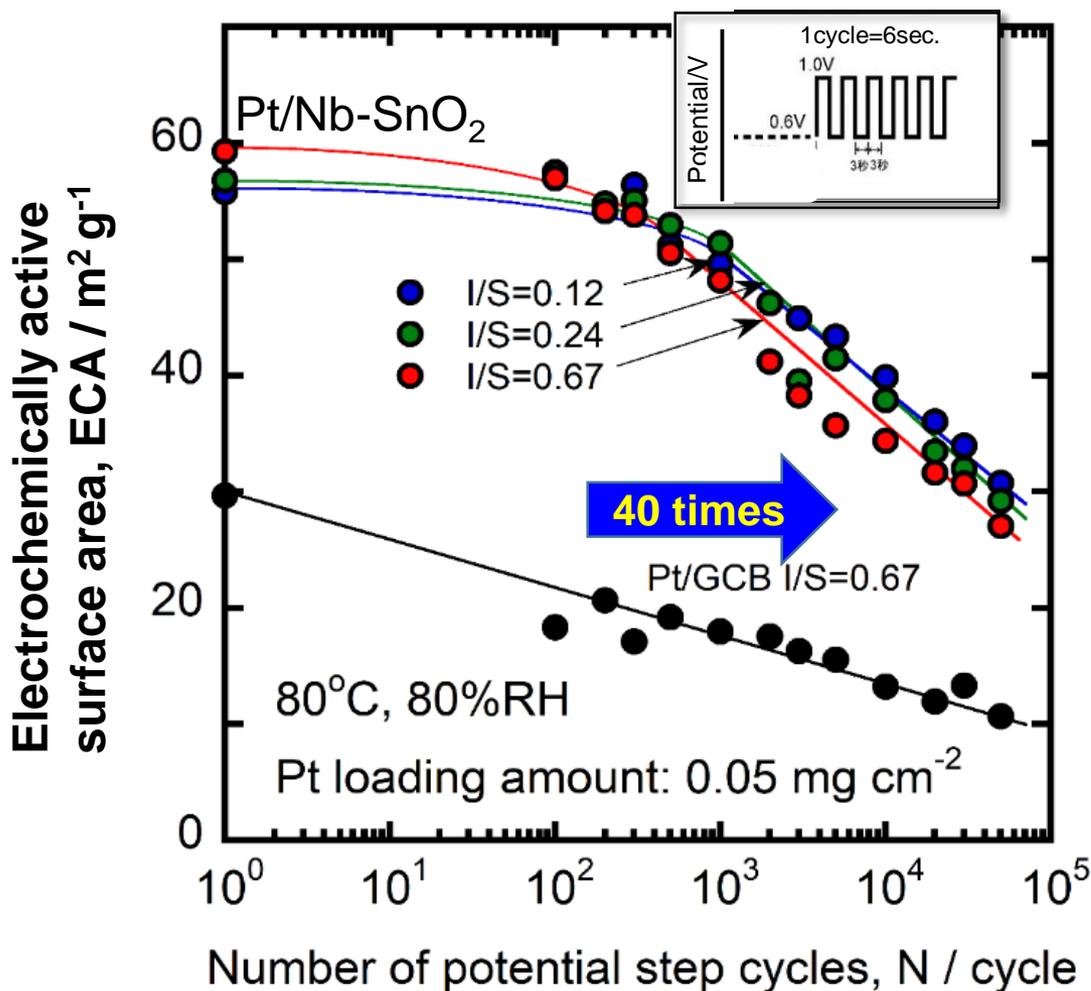




Y. Chino, K. Taniguchi, Y. Senoo, K. Kakinuma, M. Watanabe, M. Uchida, *J. Electrochem. Soc.* 162 (2015) 736.

Startup / shutdown durability of Pt/Nb-SnO<sub>2</sub> catalyst layers is superior to that of Pt/GCB catalyst layers and relies on the strong bonding between Pt and Nb-SnO<sub>2</sub>.





K. Kakinuma, R. Kobayashi, A. Iiyama, M. Uchida, *J. Electrochem. Soc.* 165 (2018) J3083.

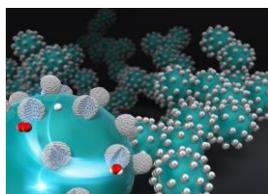
Load cycle durability of Pt/Nb-SnO<sub>2</sub> catalyst layers is also superior to that of Pt/GCB catalyst layers.

Candidate catalyst	Electronic Conductivity	Catalytic Activity	Durability	Cost
Commercial Pt/Carbon	✓	✓	✓	✓
Pt/Graphitized carbon	✓	✓	✓	✓
Pt/TiO <sub>2</sub> + Carbon nanotube	✓	✓	✓	/
Pt/TiO <sub>2</sub> -RuO <sub>2</sub>	✓	✓✓	✓✓	/
<b>Pt/Nb-SnO<sub>2</sub></b>	<b>✓✓</b>	<b>✓✓✓</b>	<b>✓✓✓</b>	<b>✓</b>

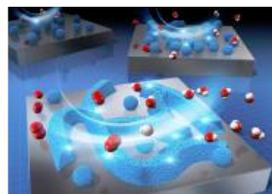
× 10  
(vs. Pt/Carbon)

× 3.2  
(vs. Pt/Carbon)

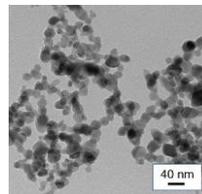
× 5000  
(vs. Pt/Carbon)



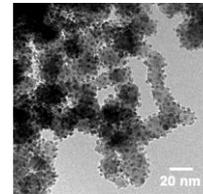
Pt nanoparticle



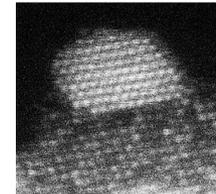
Pt nanorod



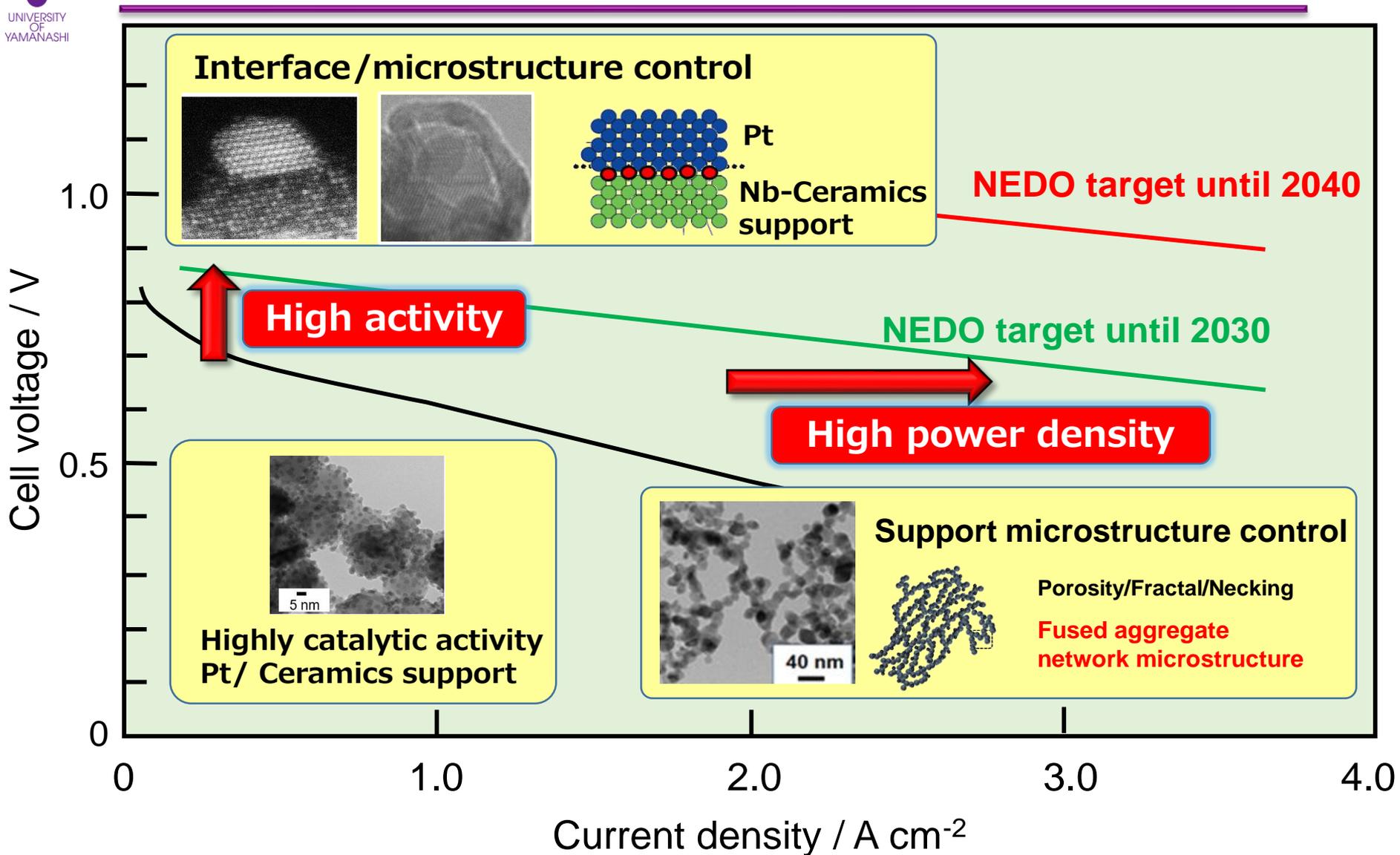
Nb support

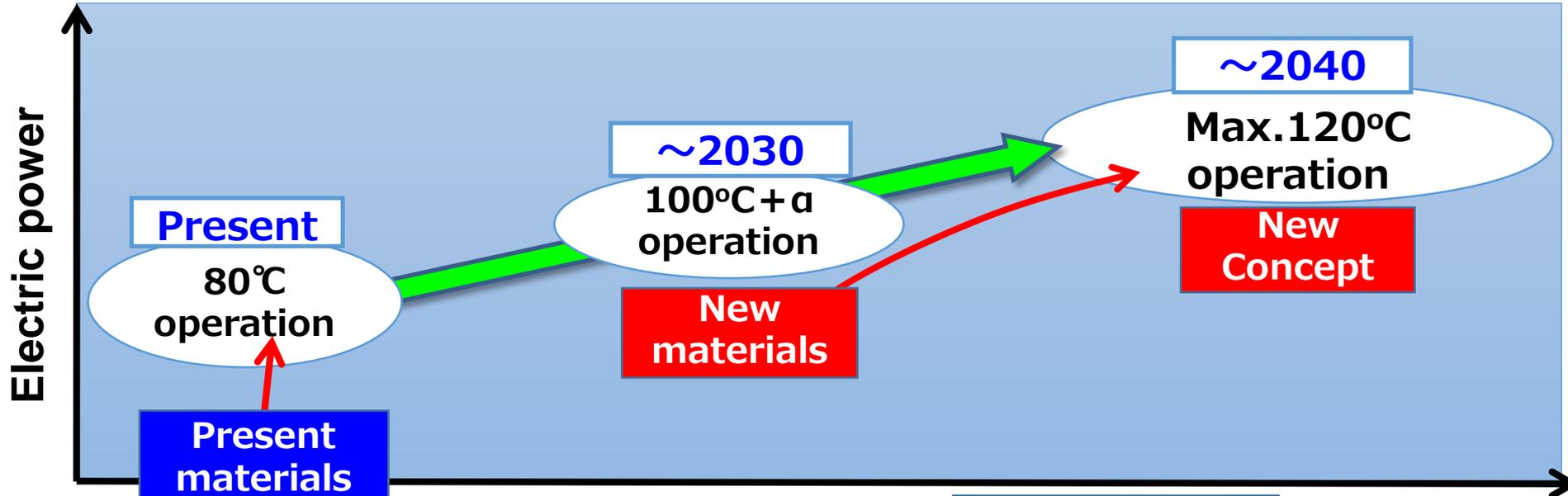


High dispersion



Pt orientation





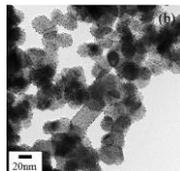
**Present concept**

Pt alloy  
Skin  
Core-shell  
Carbon  
Nafion



**New materials**  
Pt nanowire  
Fused-aggregated network nanoparticles  
Aromatic membrane with reinforcement

**New concept**  
Pt/ceramic/ionomer harmonized interface  
Platinum anti-dissolution mechanism (PADM)  
Atmospheric resistive switching mechanism (ARSM)



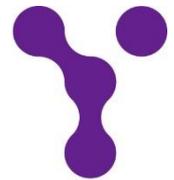
**Nb**

**Key materials**  
Pt Catalysts using Niobium supports



# Acknowledgement

This work was partially supported by funds for the New Energy and Industrial Technology Development Organization (NEDO) of Japan, and JSPS “KAKENHI” from MEXT.



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**Thank you very much  
for your kind attention !**

