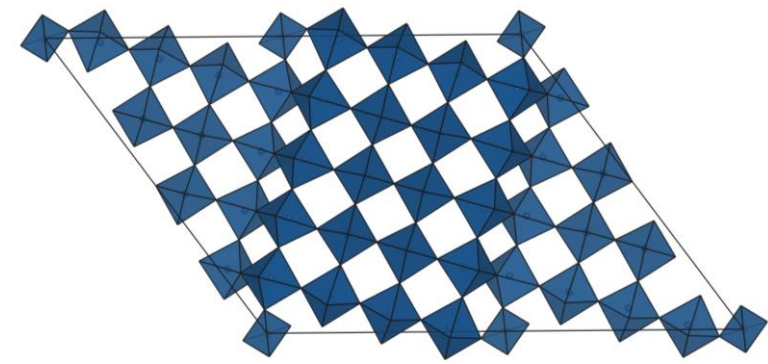
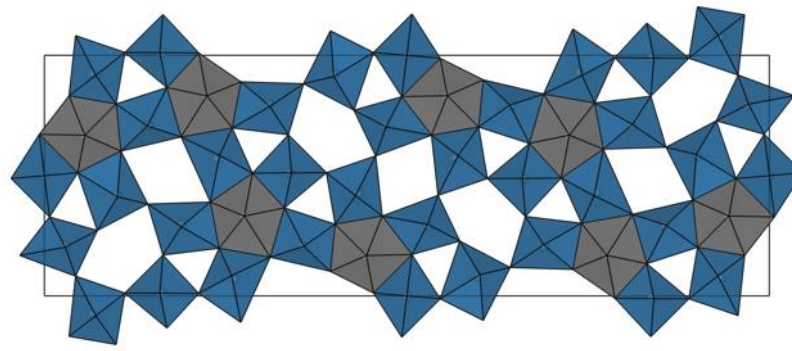




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Niobium tungsten oxides for high-rate lithium-ion energy storage

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41st Charles Hatchett Award Seminar, London

Nature **2018**, 559, 556–563.

Electrochemical energy storage

UK set to ban petrol and diesel vehicle sales from 2040

£65 million Faraday Institution for advanced batteries

Grid-scale renewables are increasing and require storage/shifting

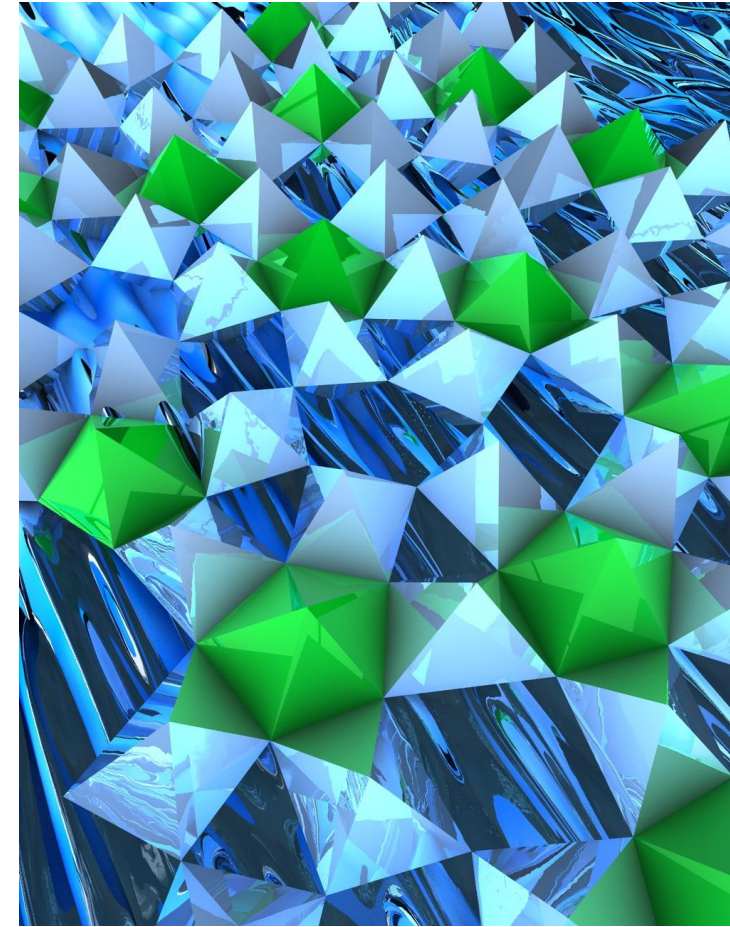
Personal electronics, power tools, internet-of-things (IoT), robotics

Lithium-ion battery market (cell level)

🔋 2018 → \$31 billion, 160 GWh

🔋 2025 → \$80 billion, 600 GWh

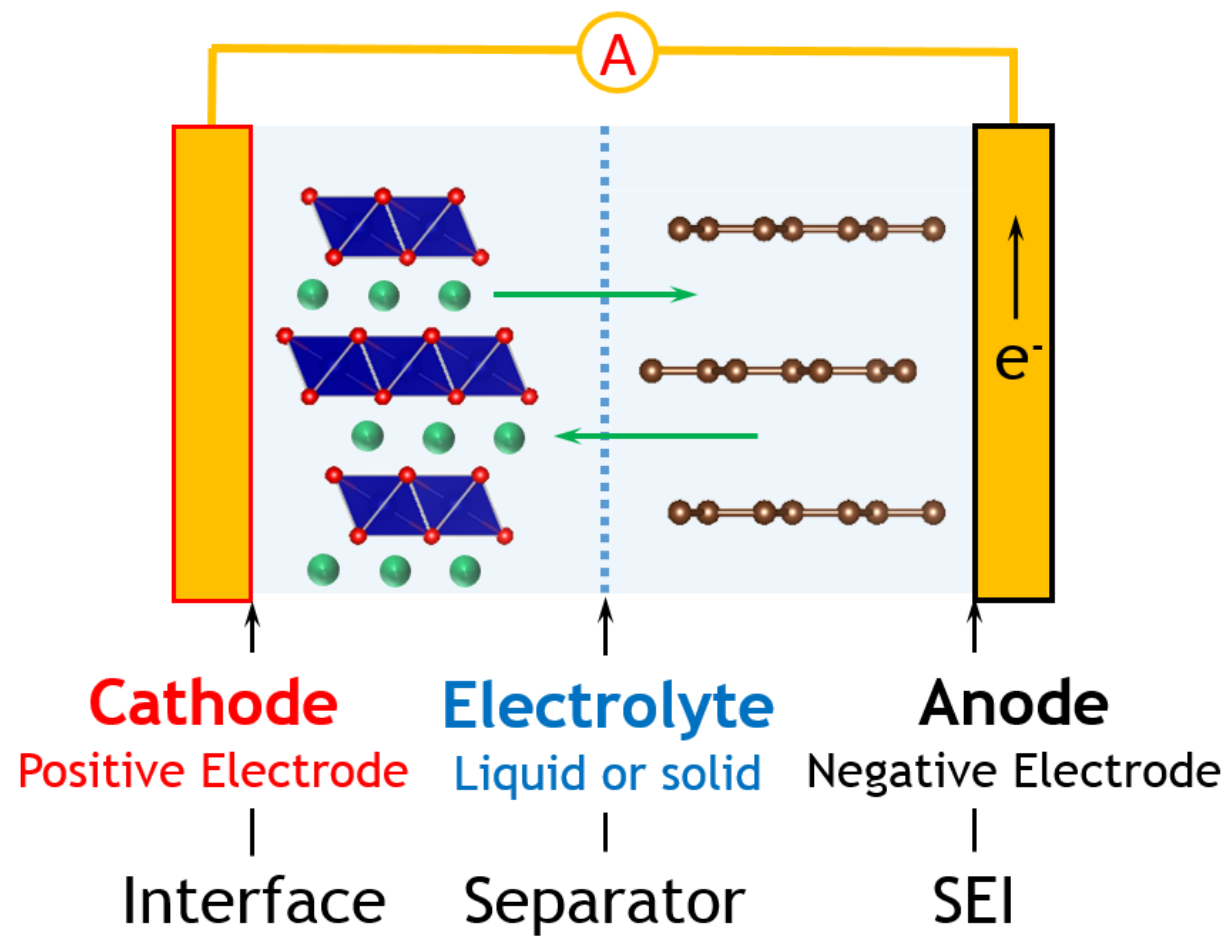
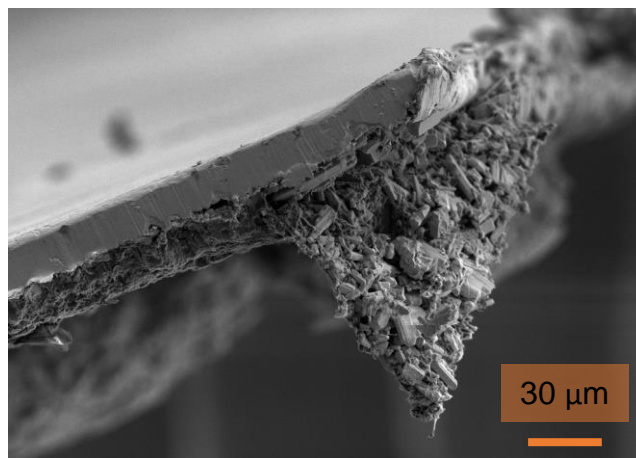
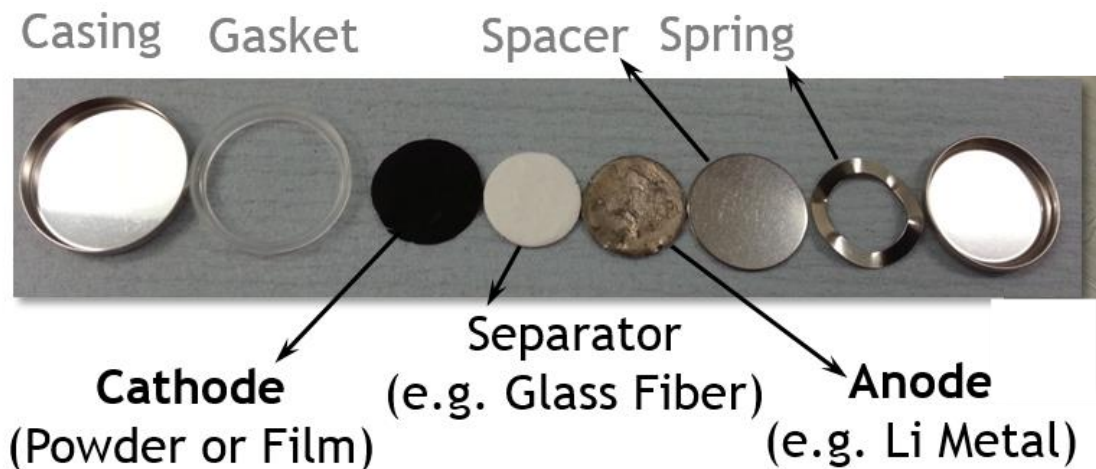
🔋 2030 → \$140 billion, 1200 GWh



Battery Applications



Lithium-ion batteries



State-of-the-art in high power anodes

Lithium titanate spinel: $\text{Li}_4\text{Ti}_5\text{O}_{12}$, LTO

Voltage vs. Li^+/Li : 1.55 V \rightarrow safety, lower energy

Max. theoretical capacity (3 Li/5 Ti):

175 $\text{mA}\cdot\text{h}\cdot\text{g}^{-1}$ (less in practice)

Long cycle life: >15,000 cycles

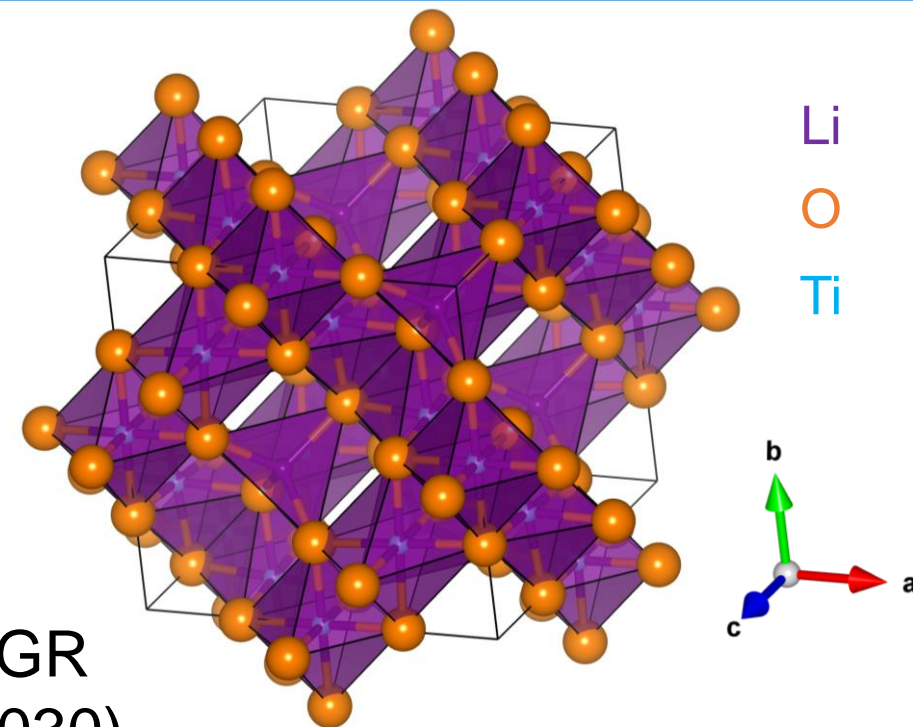
Limited Li^+ diffusion & e^- conductivity \rightarrow nanoscale

Commercial: small anode market share but 25% CAGR

4200 tons/y (2018) \rightarrow 50,000 tons/y (2030)

Improved high-rate anodes are desired for safe, long lasting, fast charging batteries

TiNb_2O_7 (Toshiba), crystallographic shear structure

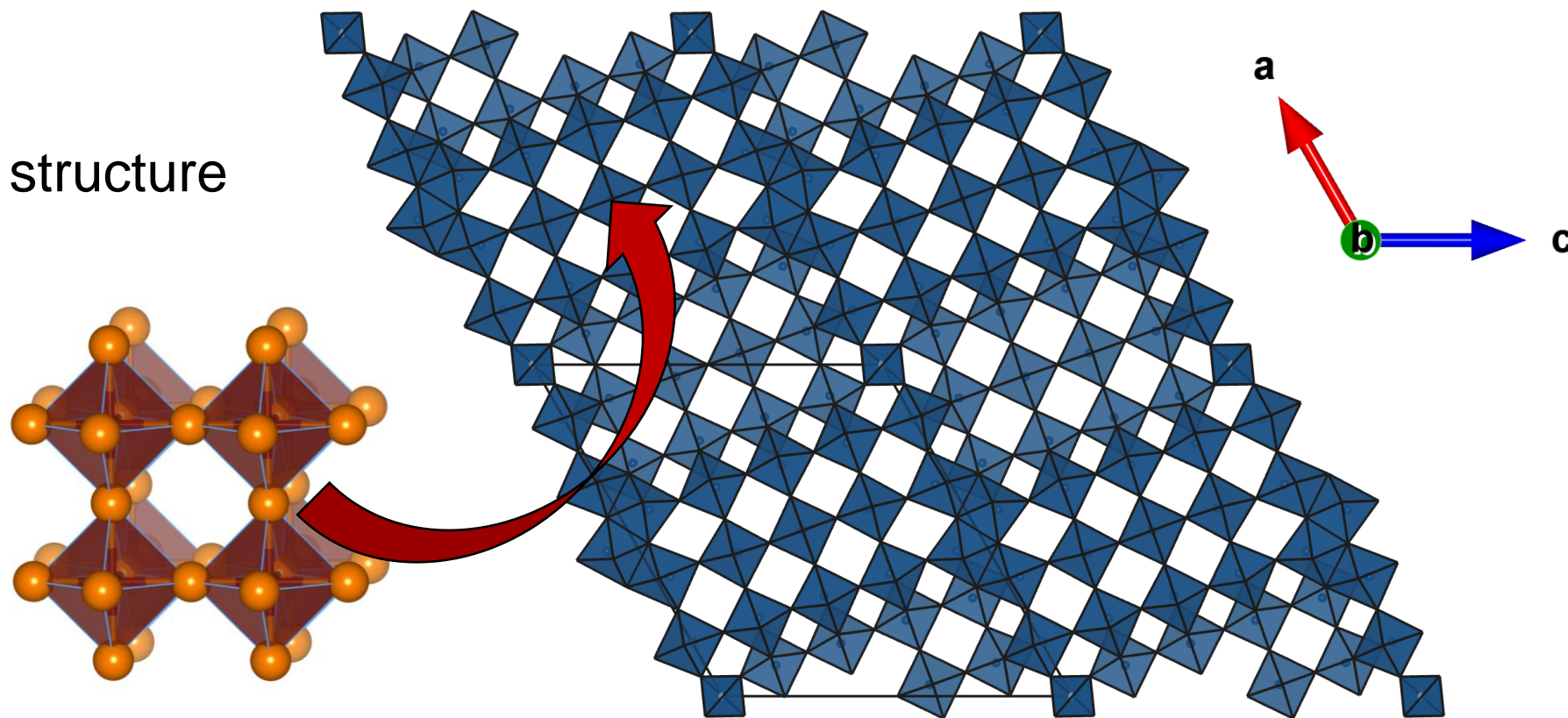


New anode materials for high power, fast charging lithium-ion batteries

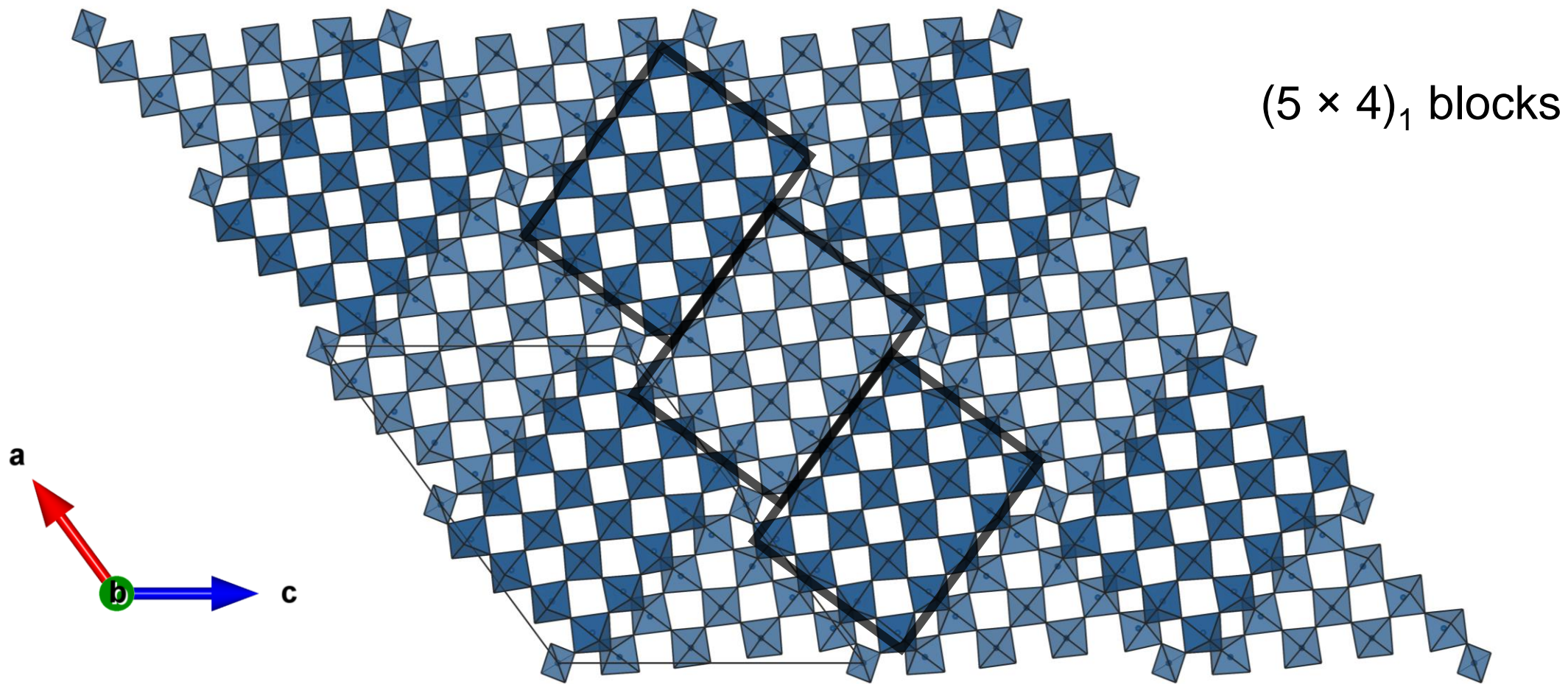
Niobium-based mixed metal oxides from lessons learnt on Nb_2O_5

H- Nb_2O_5
Wadsley–Roth
crystallographic shear structure
(4×3)₁ & (5×3)_∞

22 Ti	23 V	24 Cr
40 Zr	41 Nb	42 Mo
72 Hf	73 Ta	74 W

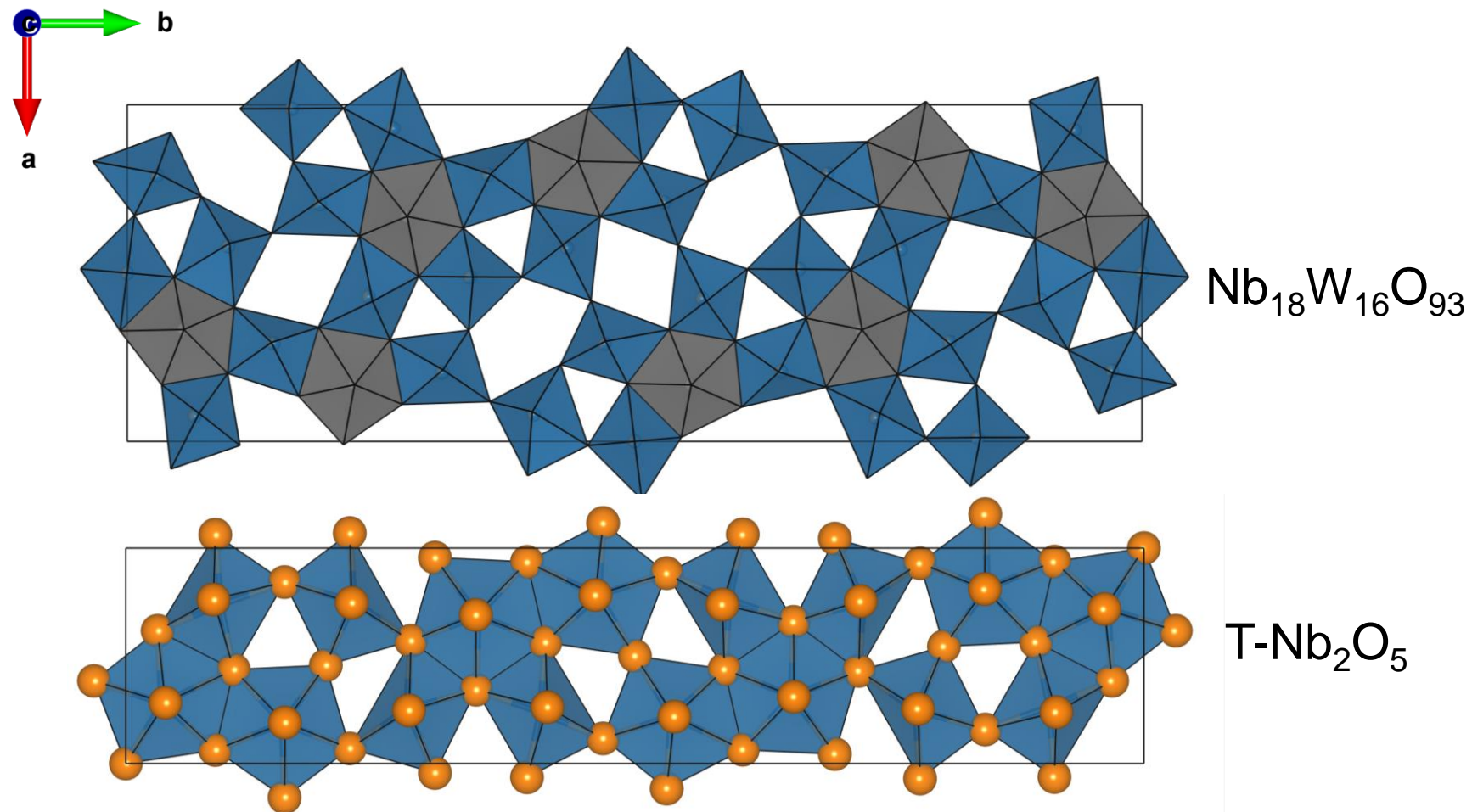


Nb₁₆W₅O₅₅ crystal structure



New anode materials for high power, fast charging lithium-ion batteries

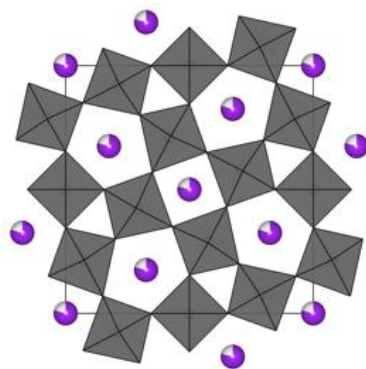
Niobium-based mixed metal oxides from lessons learnt on Nb_2O_5



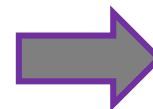
Nb₁₈W₁₆O₉₃ crystal structure

Nb₁₈W₁₆O₉₃
a tetragonal tungsten
bronze (TTB) derivative

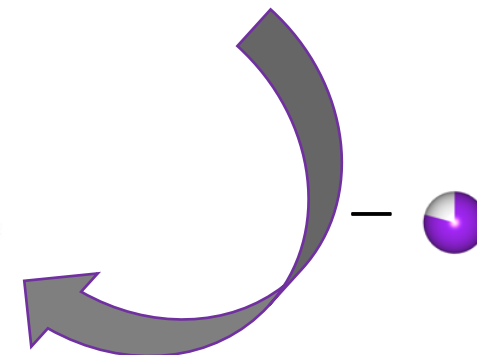
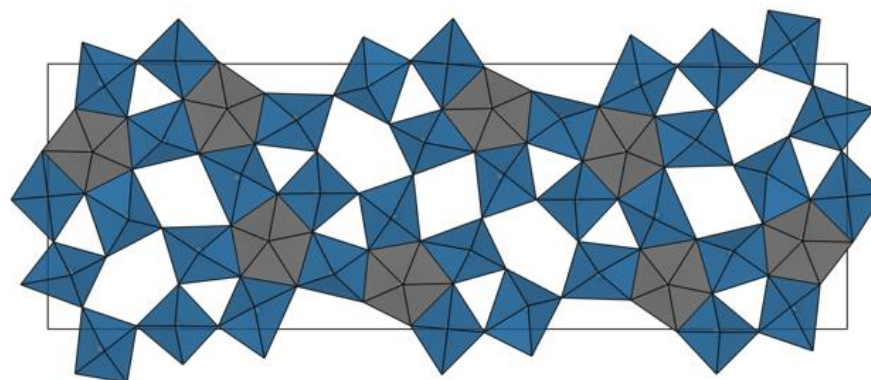
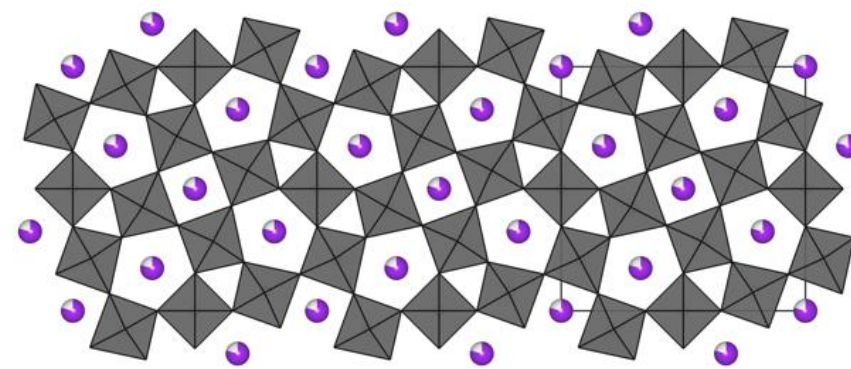
TTB (K_xWO₃)



× 3

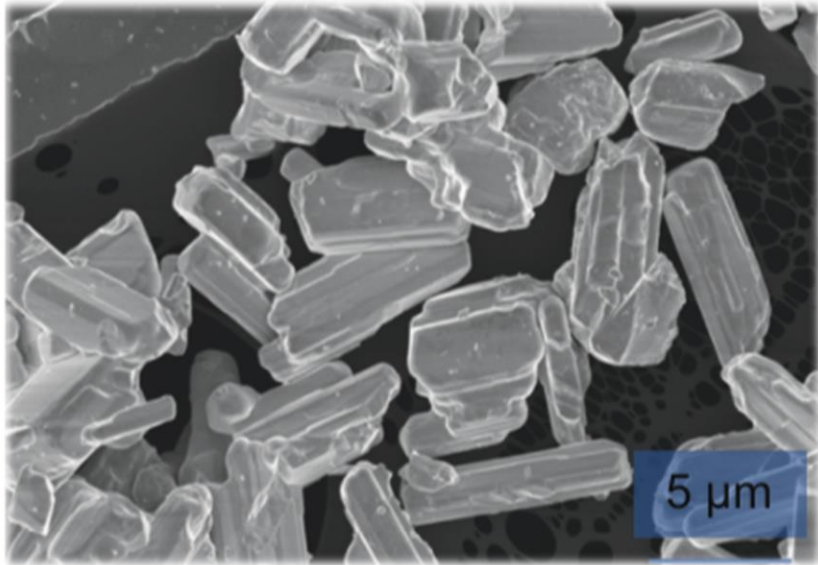


TTB supercell

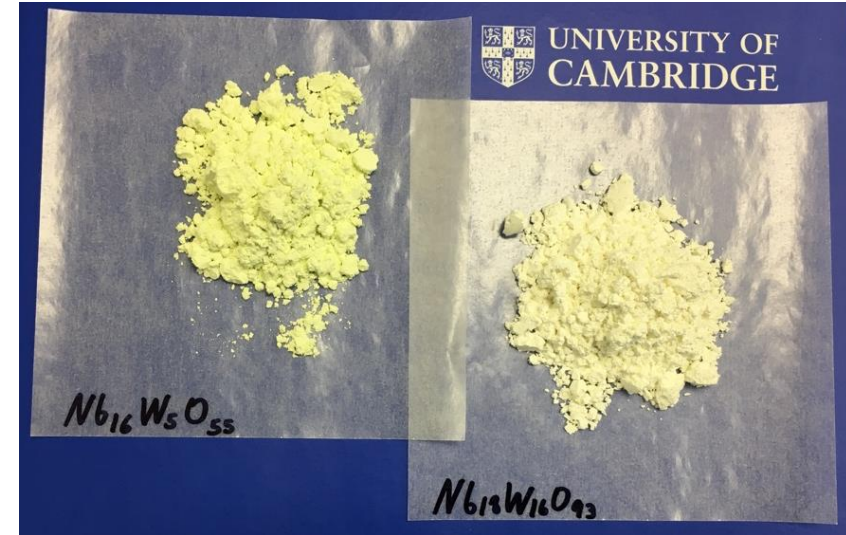
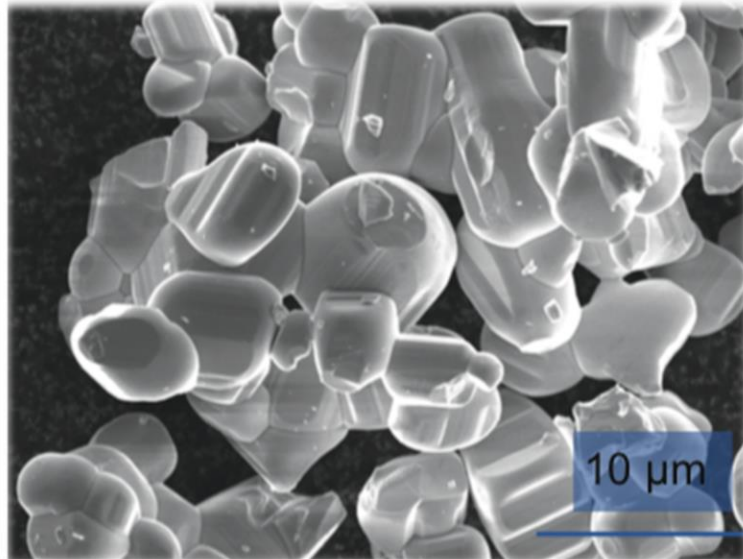


Nb₁₈W₁₆O₉₃ - distorted TTB superstructure

Micrometer-scale bulk particle morphology (for high rates??)



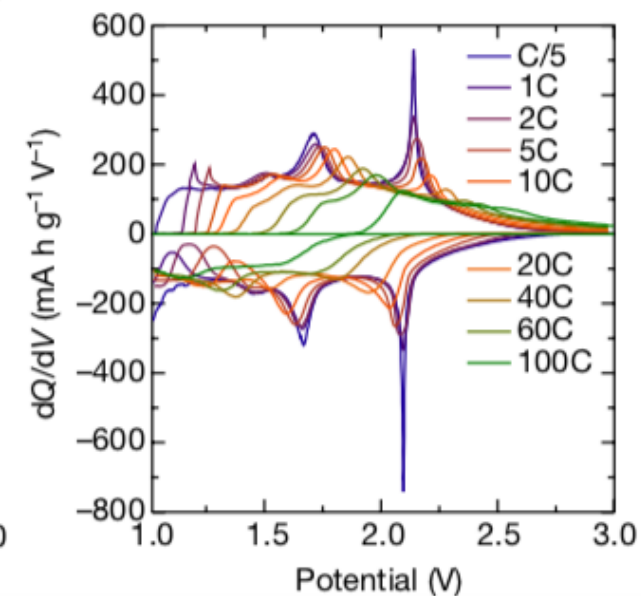
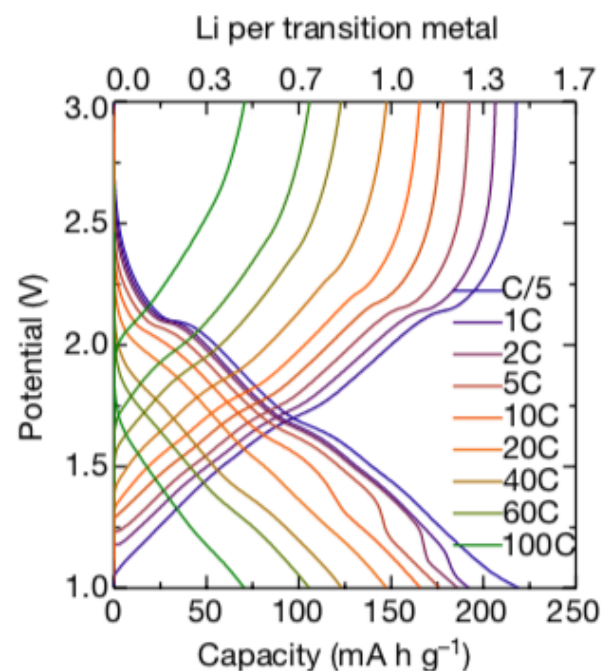
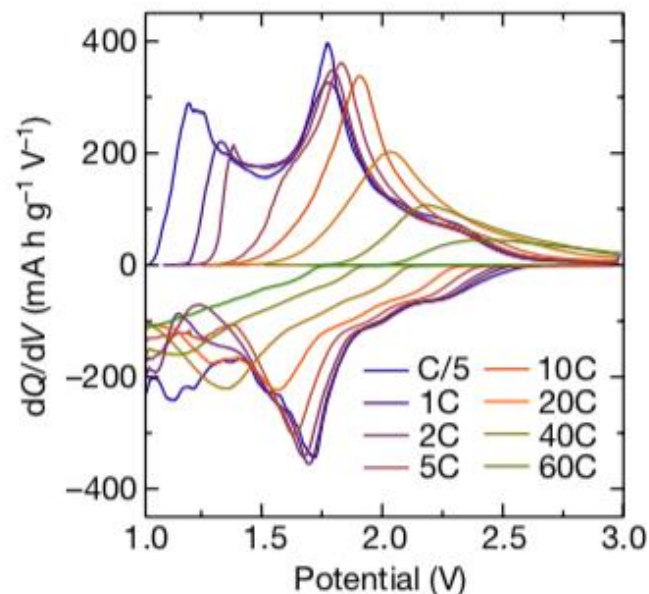
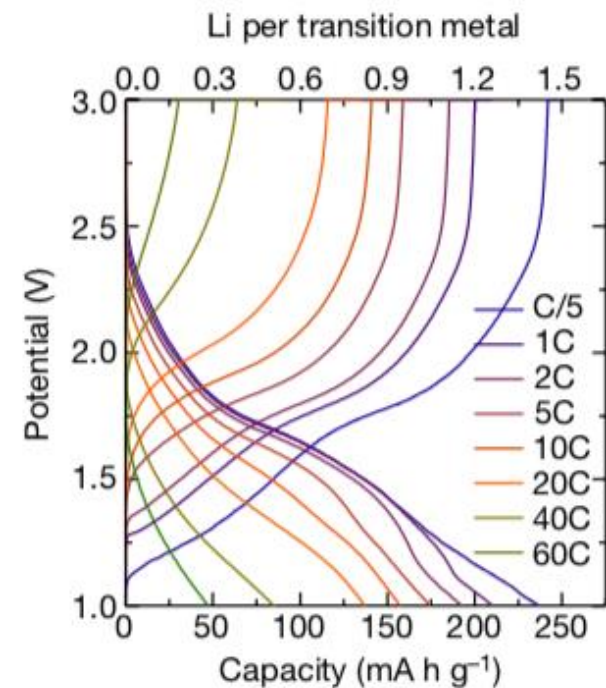
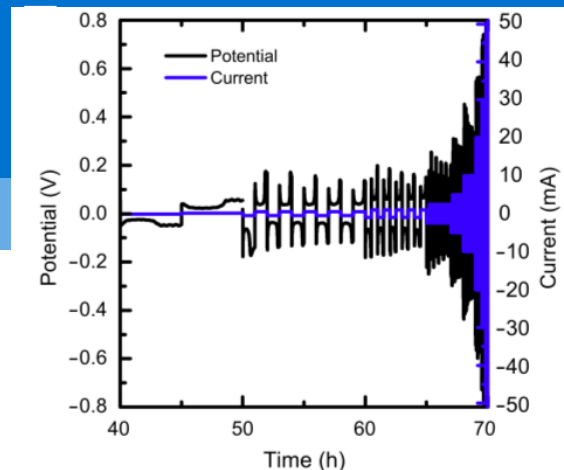
Electrode manufacturing
Standard powder mixing
Standard slurry coating



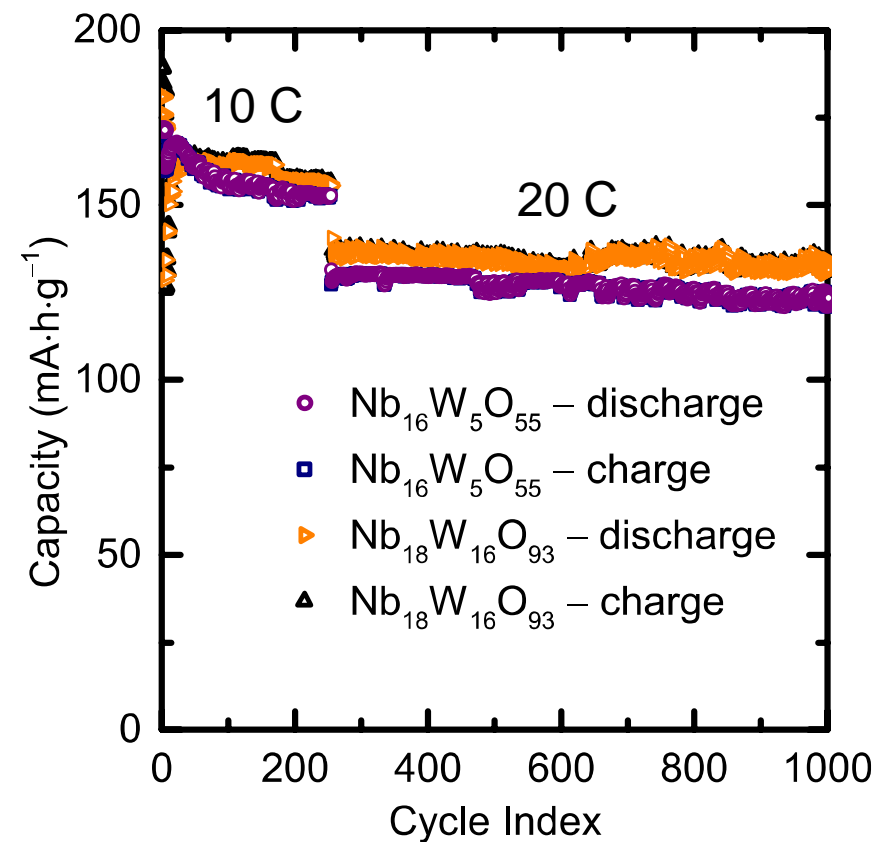
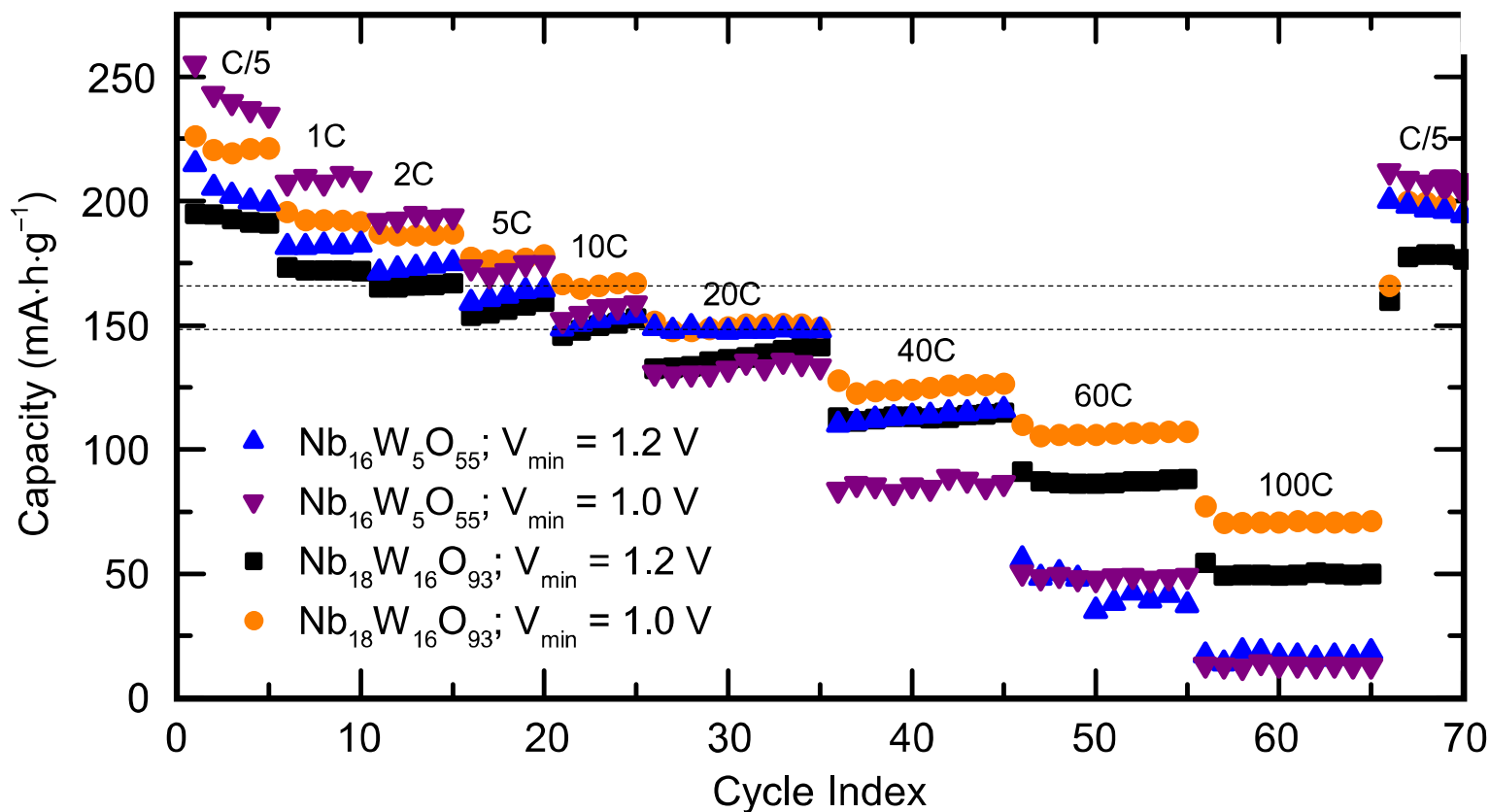
Material synthesis
Scalable
Low manufacturing cost
(Li-free synthesis)

Battery performance
Low surface area = low reactivity
→ long cycle life, high safety

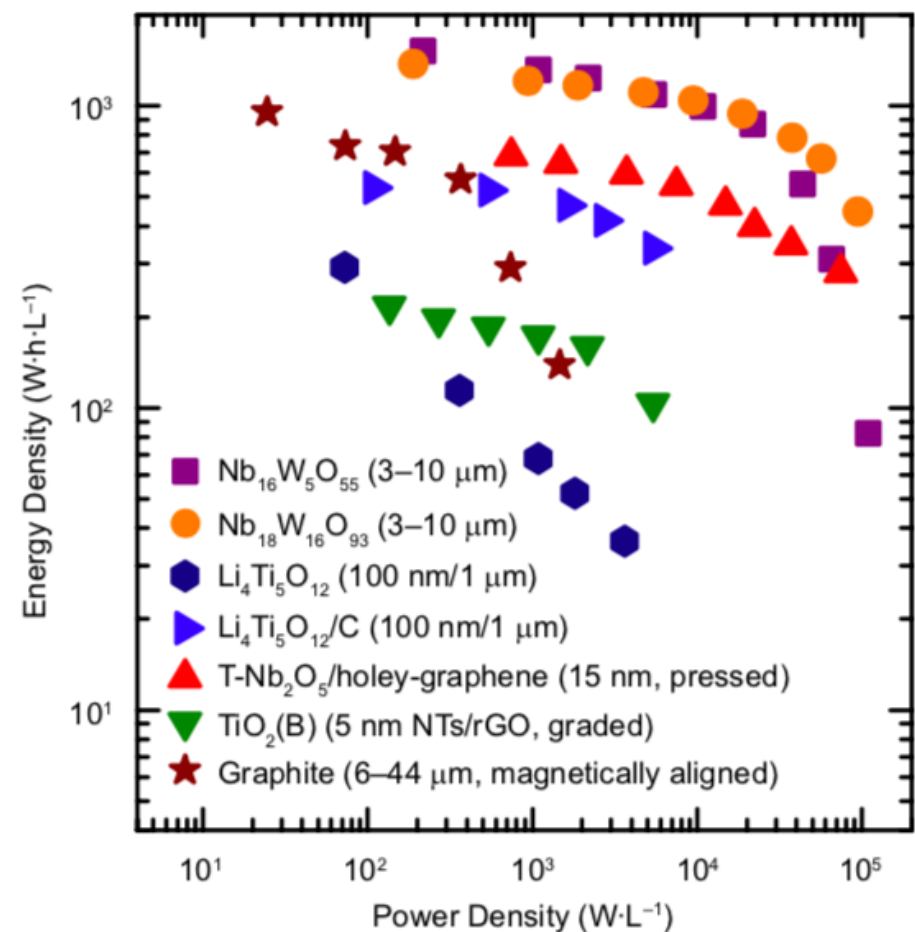
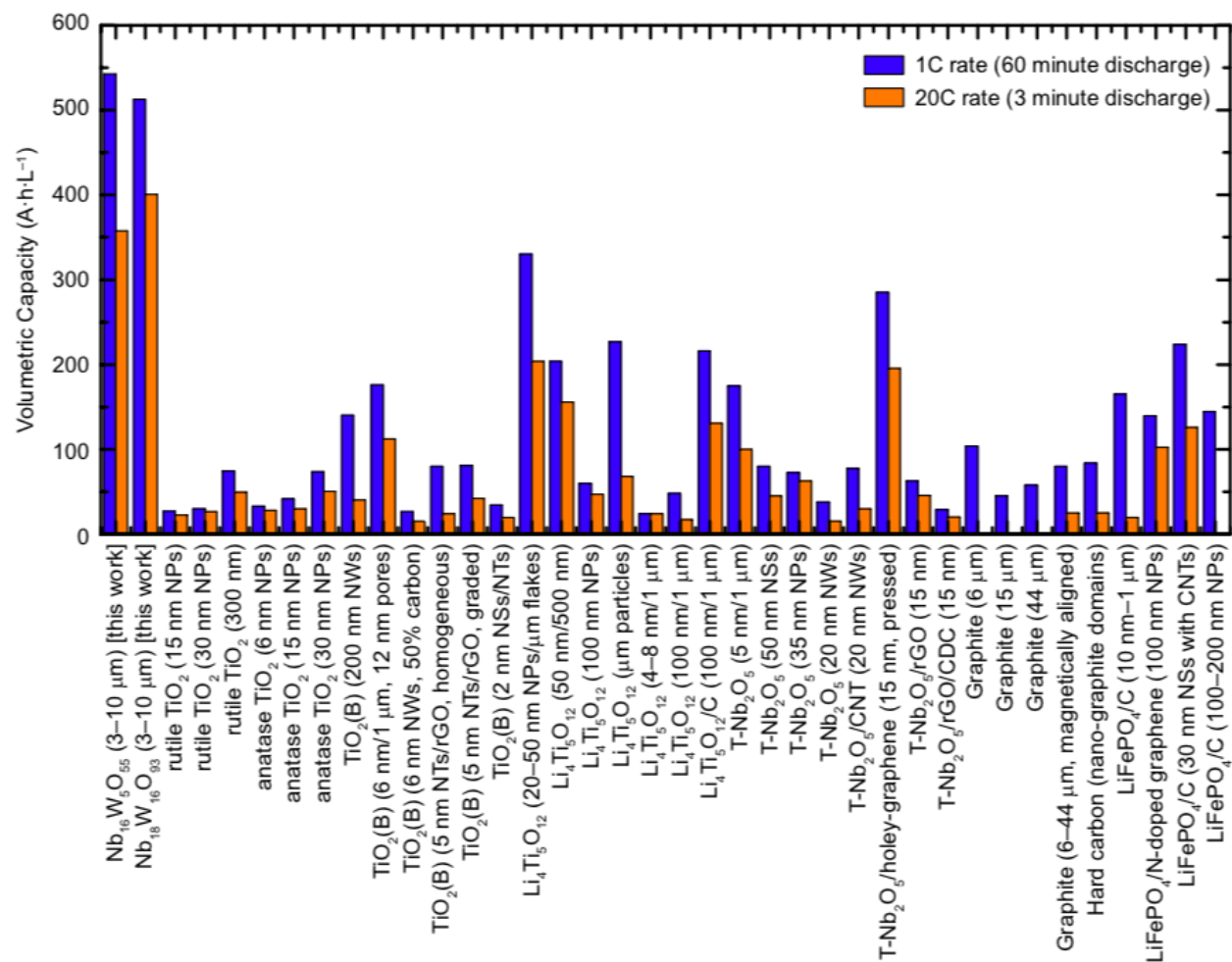
Niobium tungsten oxide electrochemistry



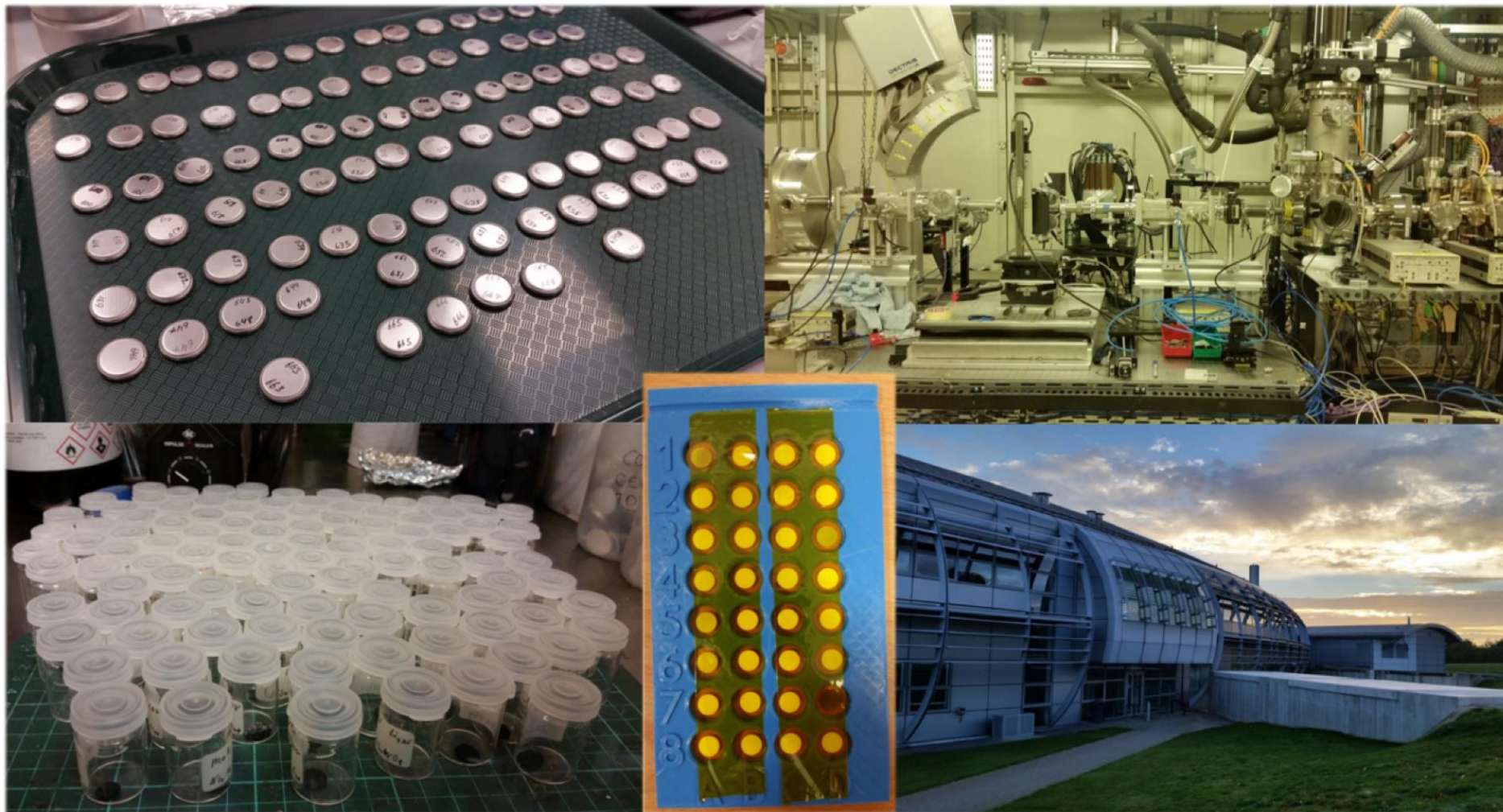
Niobium tungsten oxide electrochemistry



Niobium tungsten oxide electrochemistry



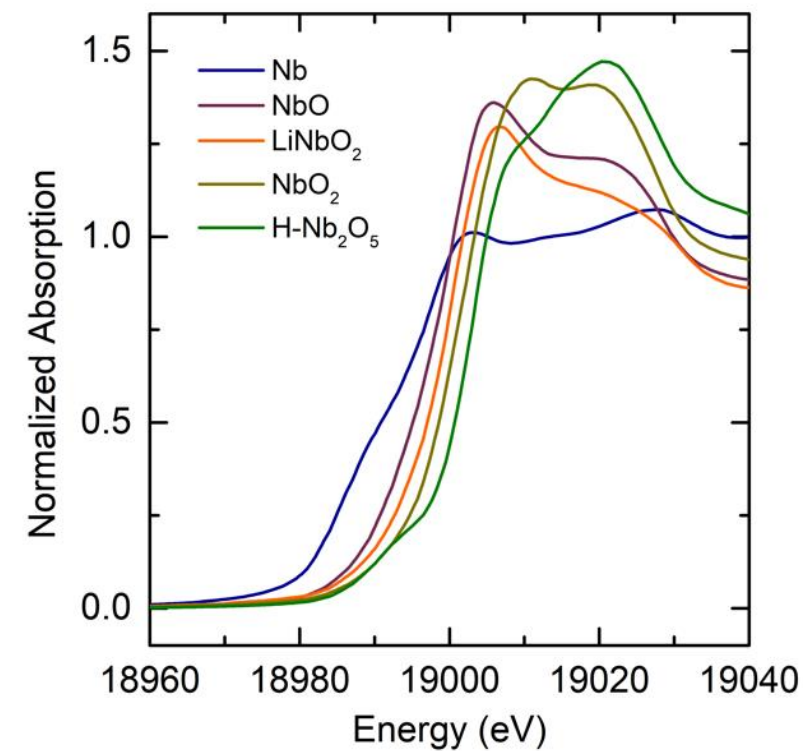
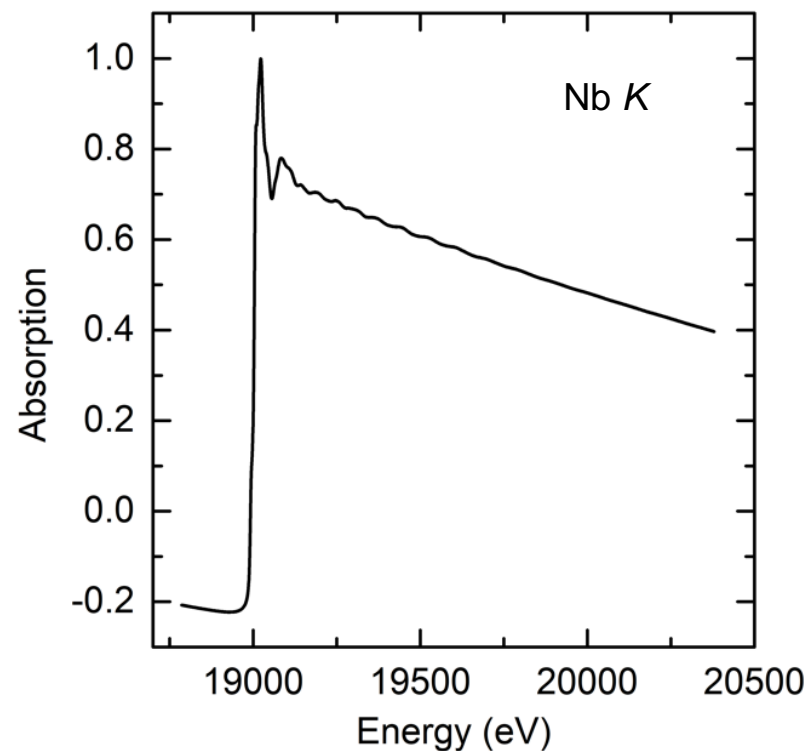
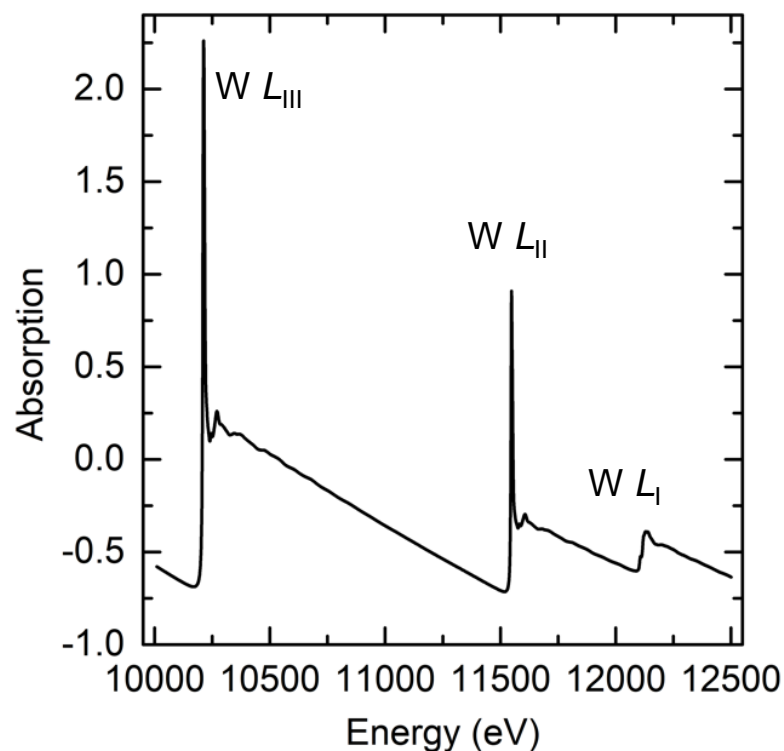
Chemical and structural insights from synchrotron X-rays



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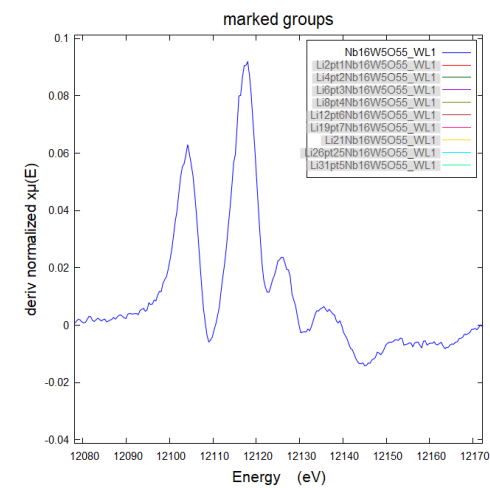
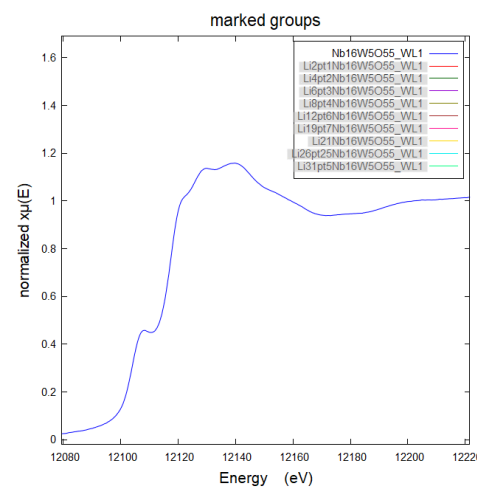
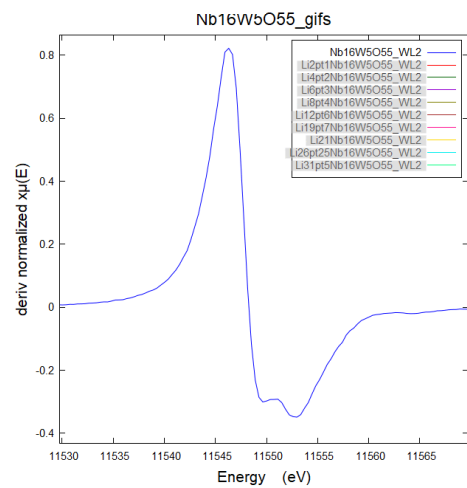
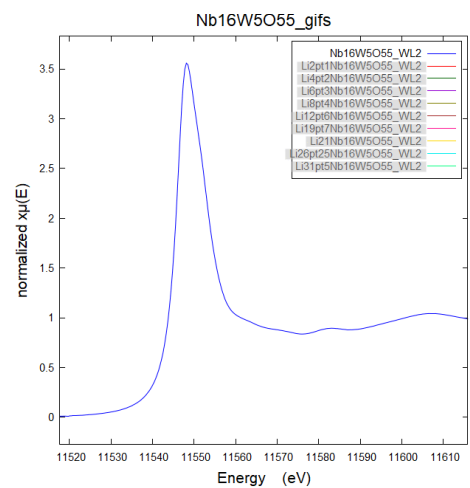
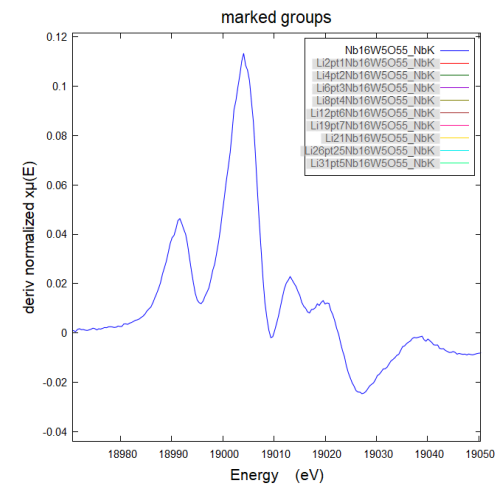
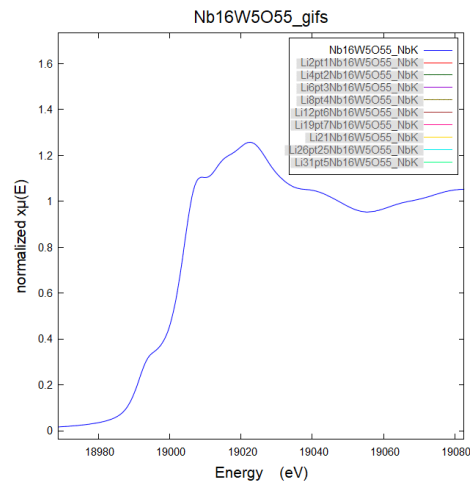
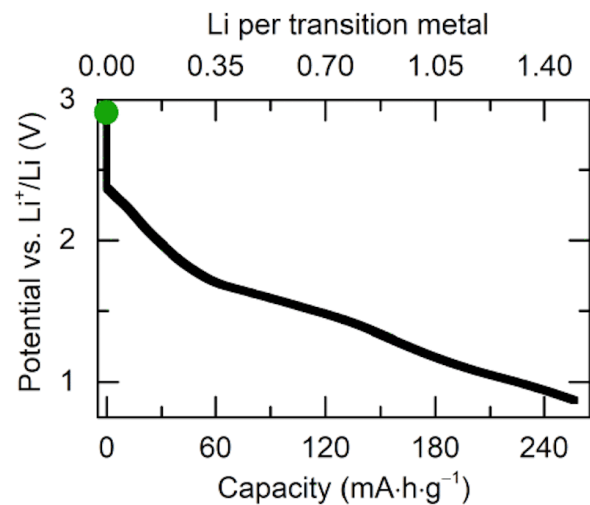
Diamond Light Source, Beamline B18
Principal beamline scientist: Giannantonio Cibin

Multi-edge X-ray absorption spectroscopy

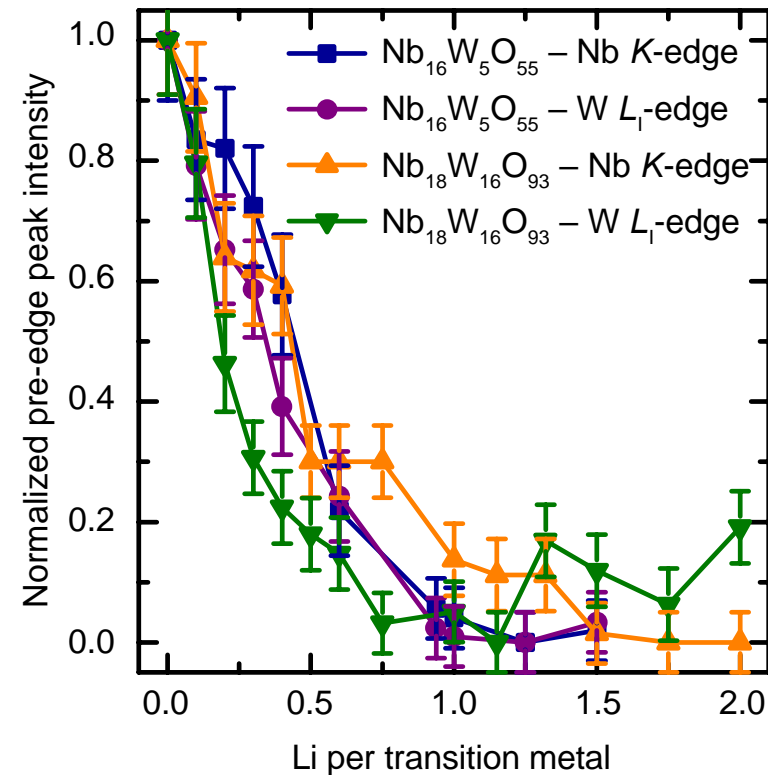
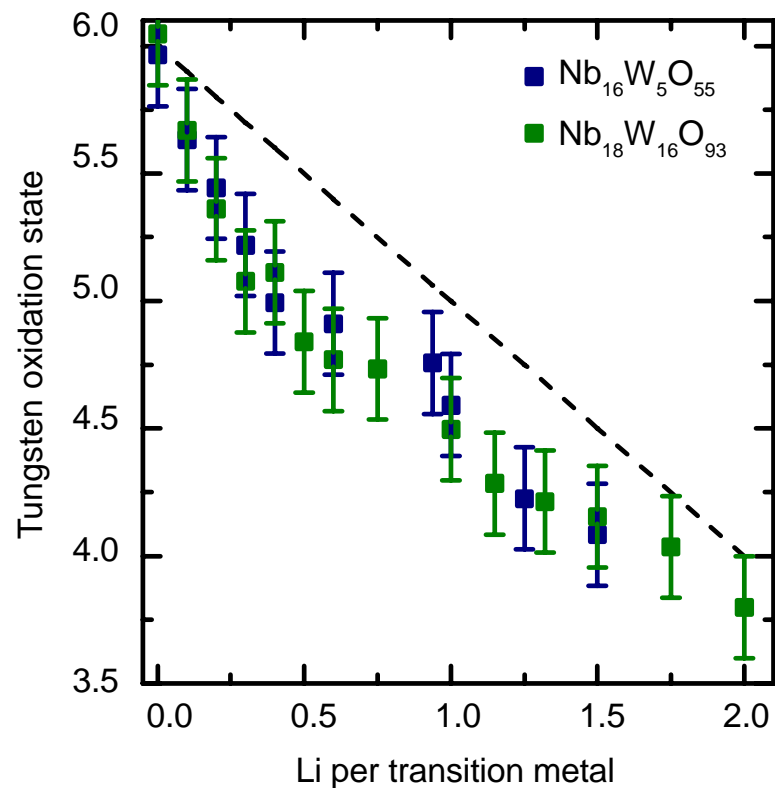
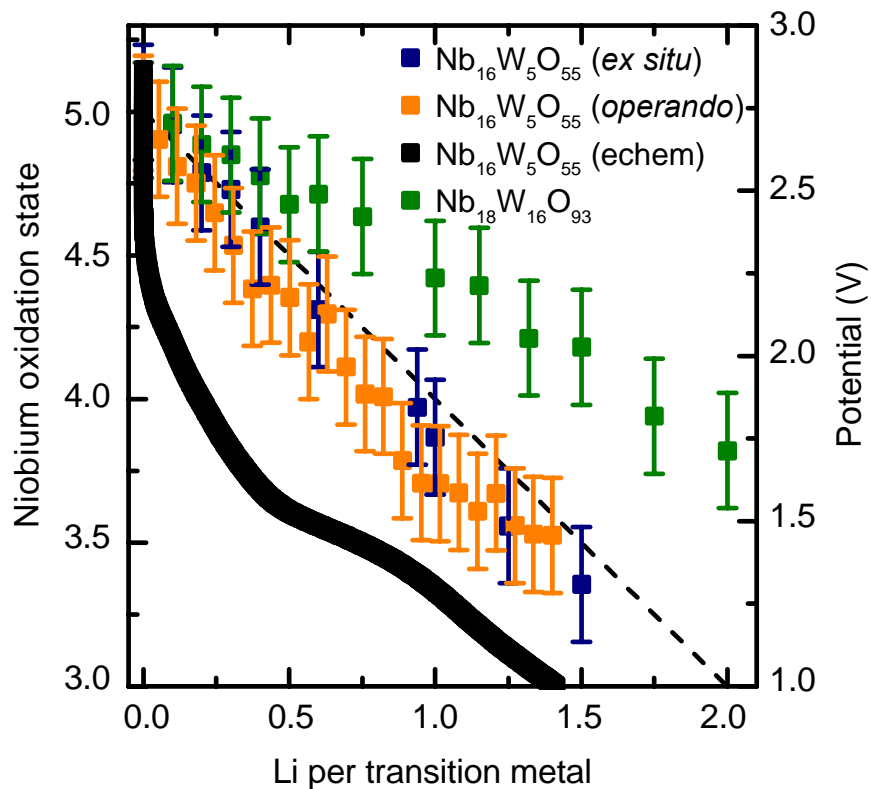


XAS: Element specific, sensitive to bulk, electronic and atomic probe

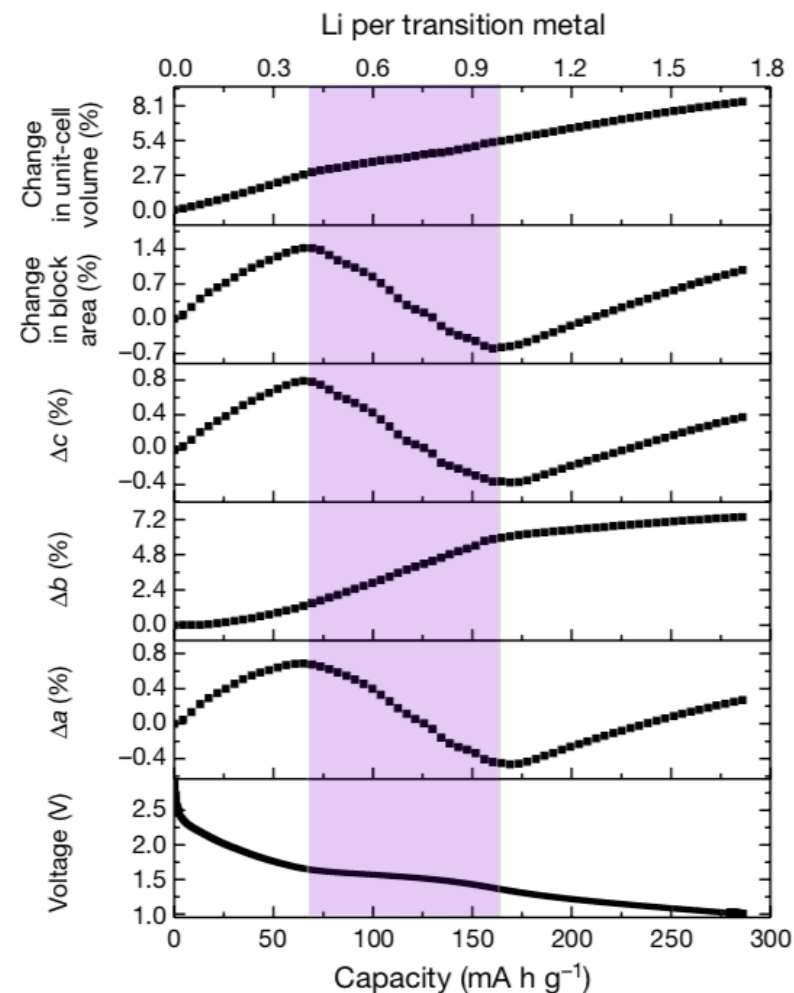
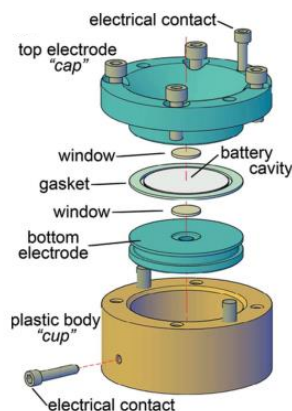
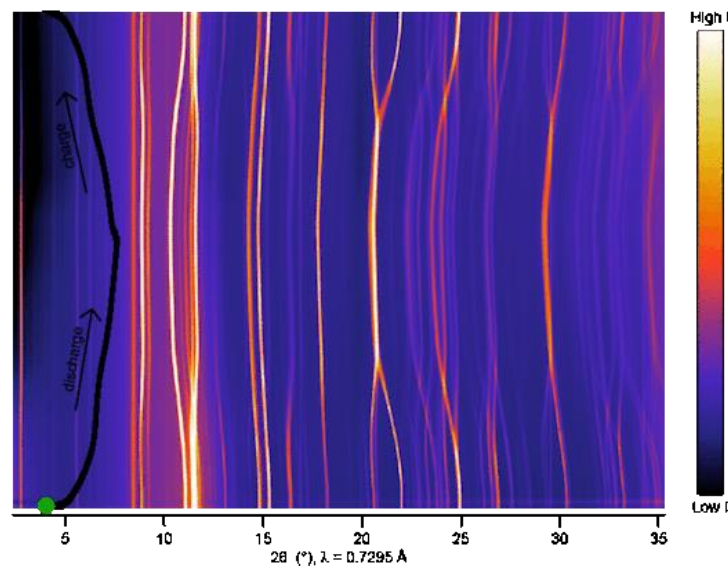
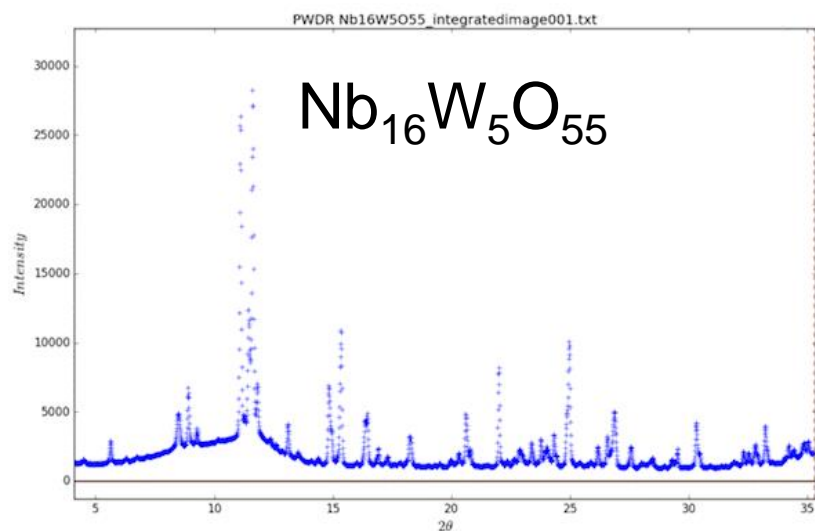
Nb₁₆W₅O₅₅ XAS @ Nb K, W L_{II}, W L_I edges



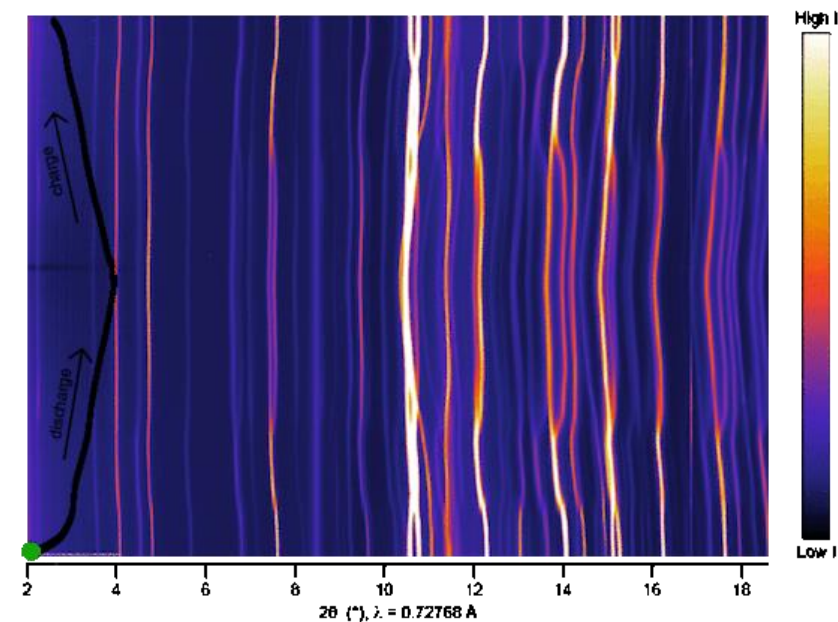
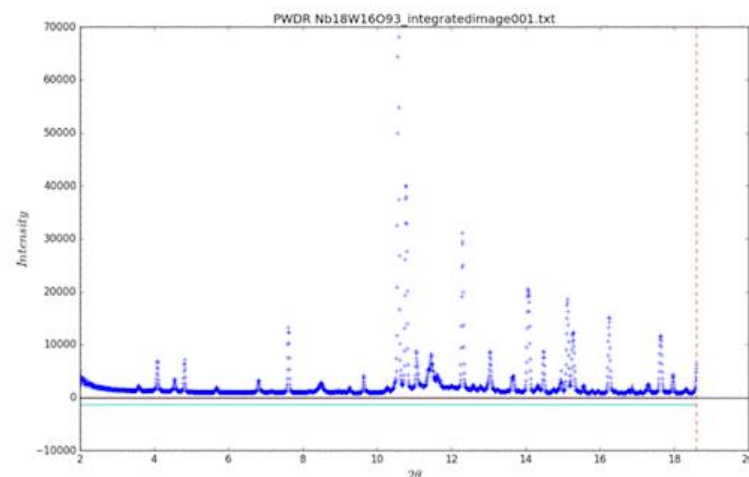
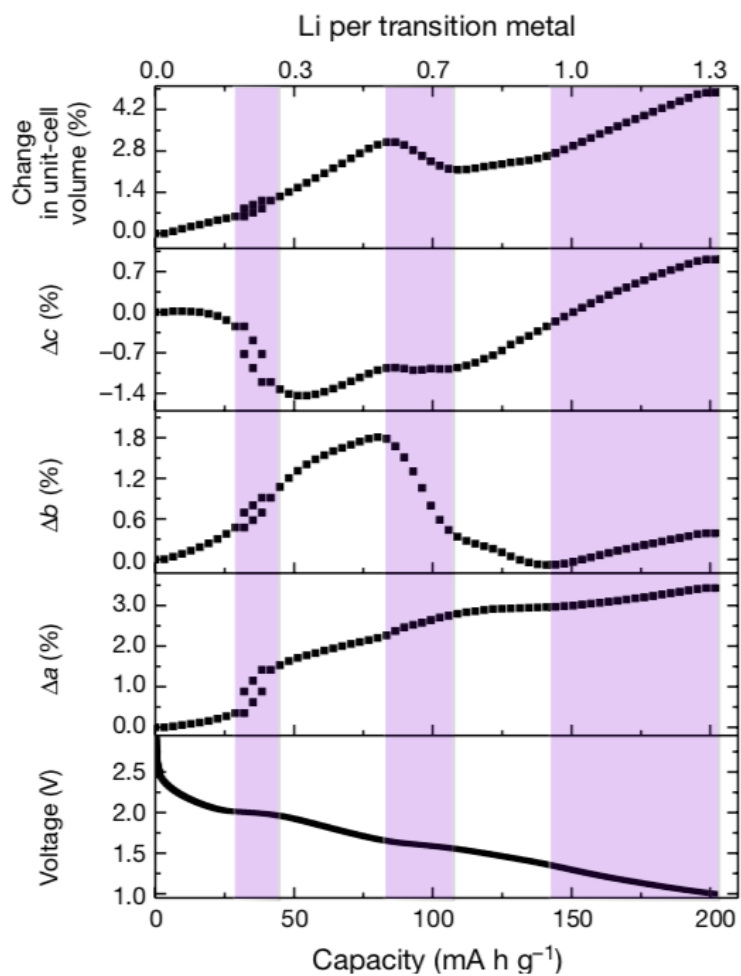
Multi-electron Redox at Nb and W



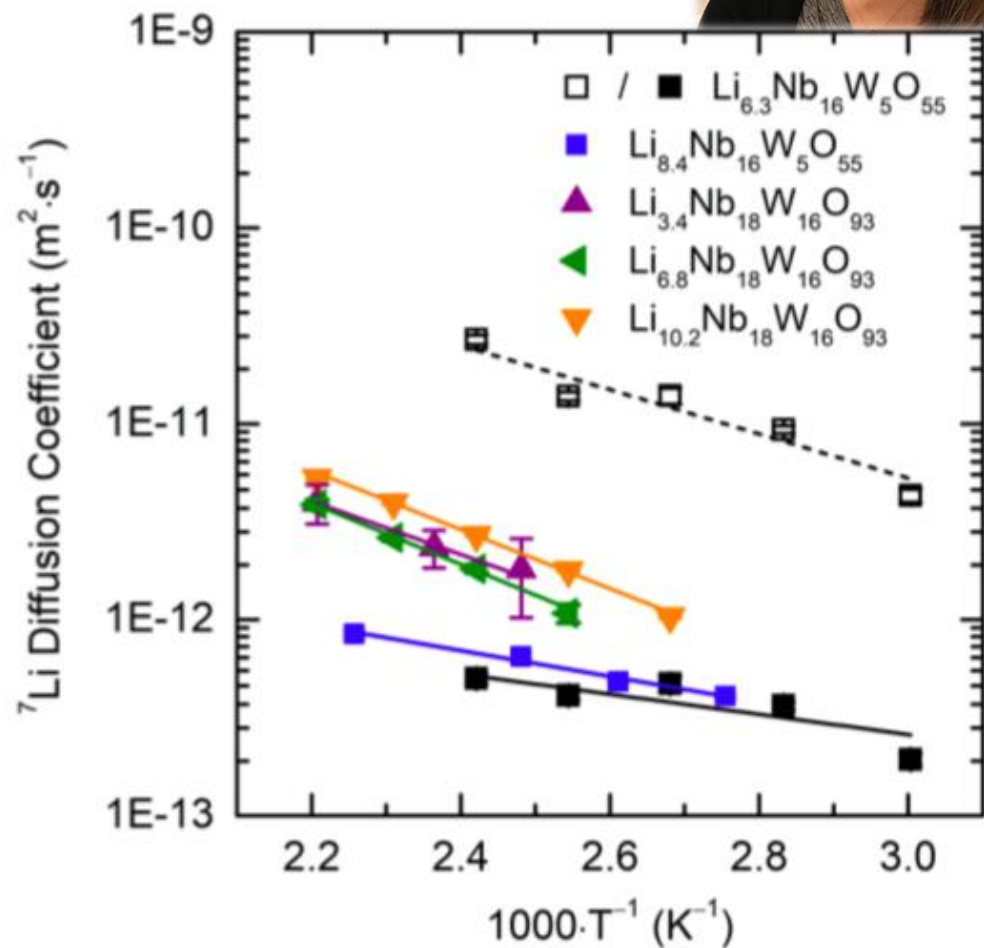
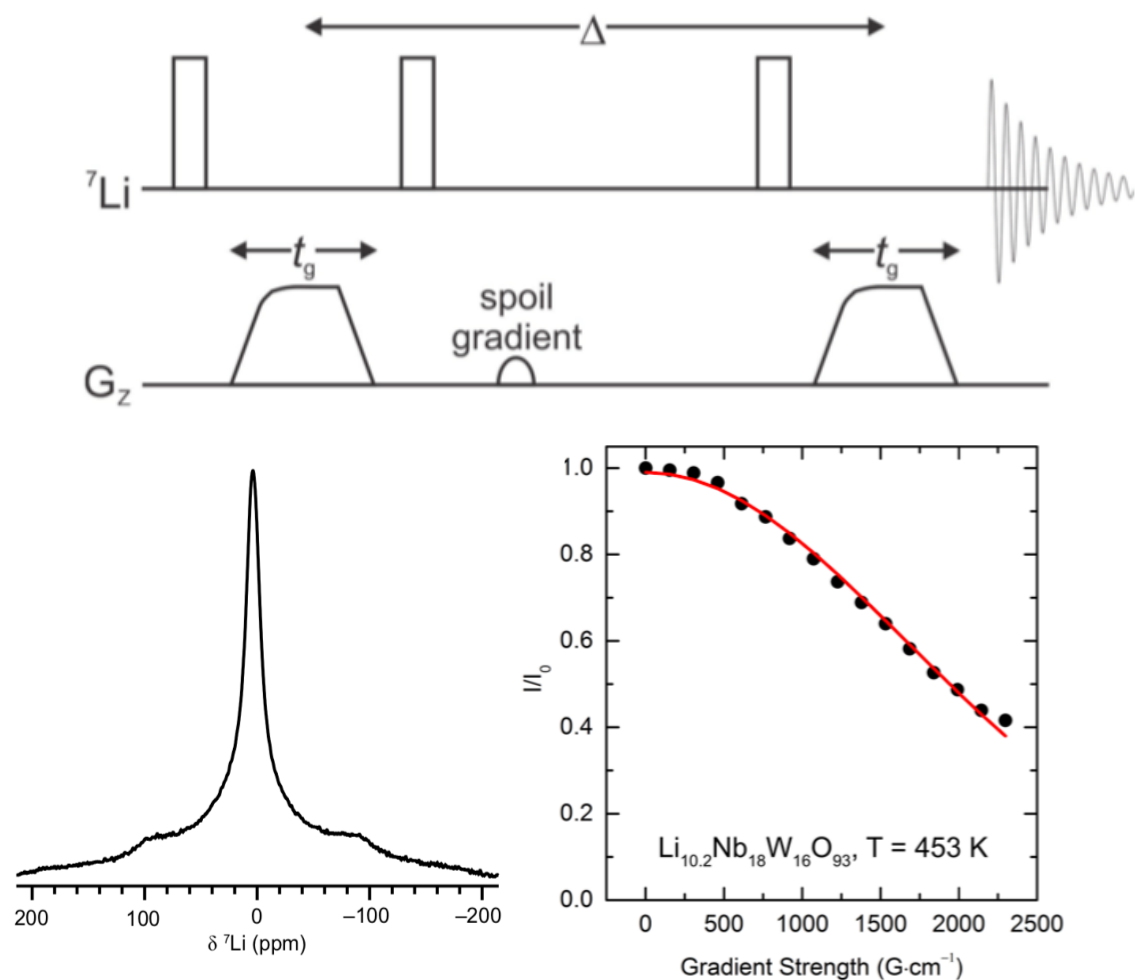
Operando high-rate structure evolution from synchrotron diffraction



Operando high-rate structure evolution from synchrotron diffraction

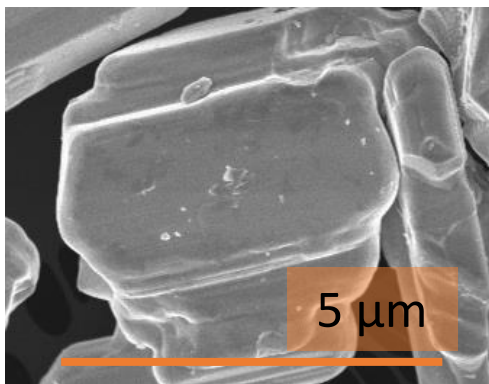


Pulsed field gradient NMR Spectroscopy



Putting diffusion coefficients into context

	Diffusion Length (μm)		
D_{Li} ($\text{m}^2\cdot\text{s}^{-1}$)	1C (3600 s)	20C (180 s)	60C (60 s)
1.0×10^{-12}	150	33	19
1.0×10^{-14}	15	3.3	1.9
1.0×10^{-16}	1.5	0.33	0.19
1.0×10^{-18}	0.15	0.033	0.019
1.0×10^{-20}	0.015	0.0033	0.0019

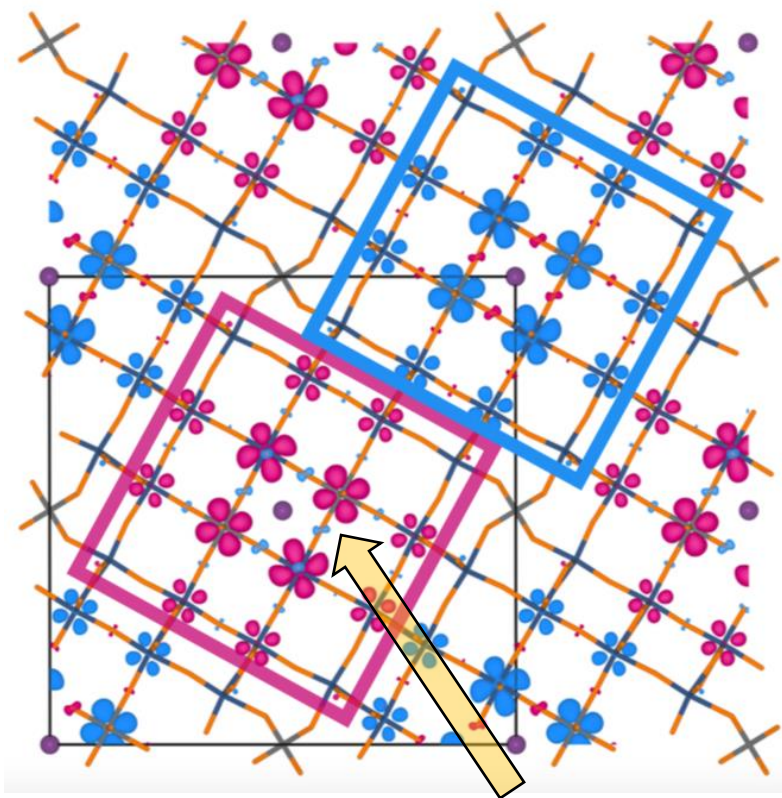


Liquid electrolytes are 10^{-10} – 10^{-12} $\text{m}^2\cdot\text{s}^{-1}$

Compound	Structure Type	D_{Li} ($\text{m}^2\cdot\text{s}^{-1}$)	T (K)	Technique	Reference
$\text{Li}_{10}\text{GeP}_2\text{S}_{12}$, Li_7GePS_8 , $\text{Li}_{10}\text{SnP}_2\text{S}_{12}$, $\text{Li}_7\text{P}_3\text{S}_{11}$, & $\text{Li}_{11}\text{Si}_2\text{PS}_{12}$	Thio-LISICON	$1\text{--}5 \times 10^{-12}$	298	PFG NMR	Kuhn et al. (2013), (2014), Hayamizu et al. (2013)
$\beta\text{-Li}_3\text{PS}_4$	Thio-LISICON	5.4×10^{-13}	373	PFG NMR	Gobet et al.
$\text{Li}_{0.6}[\text{Li}_{0.2}\text{Sn}_{0.8}\text{S}_2]$	Layered (O1)	$2\text{--}20 \times 10^{-12}$	298	PFG NMR	Holzmann et al.
$\text{Li}_{1.5}\text{Al}_{0.5}\text{Ge}_{1.5}(\text{PO}_4)_3$	NASICON	2.9×10^{-13}	311	PFG NMR	Hayamizu et al.
$\text{Li}_{6.6}\text{La}_3\text{Zr}_{1.6}\text{Ta}_{0.4}\text{O}_{12}$	Garnet	3.5×10^{-13}	353	PFG NMR	Hayamizu et al.
Graphite (Stage I)	Graphite	$1\text{--}2 \times 10^{-15}$	298	NMR relaxn.	Langer et al.
$\text{Li}_4\text{Ti}_5\text{O}_{12}$	Spinel	3.2×10^{-15}	298	μ^+ -SR	Sugiyama et al.
LiMn_2O_4	Spinel	1×10^{-20}	350	NMR relaxn.	Verhoevenm et al.

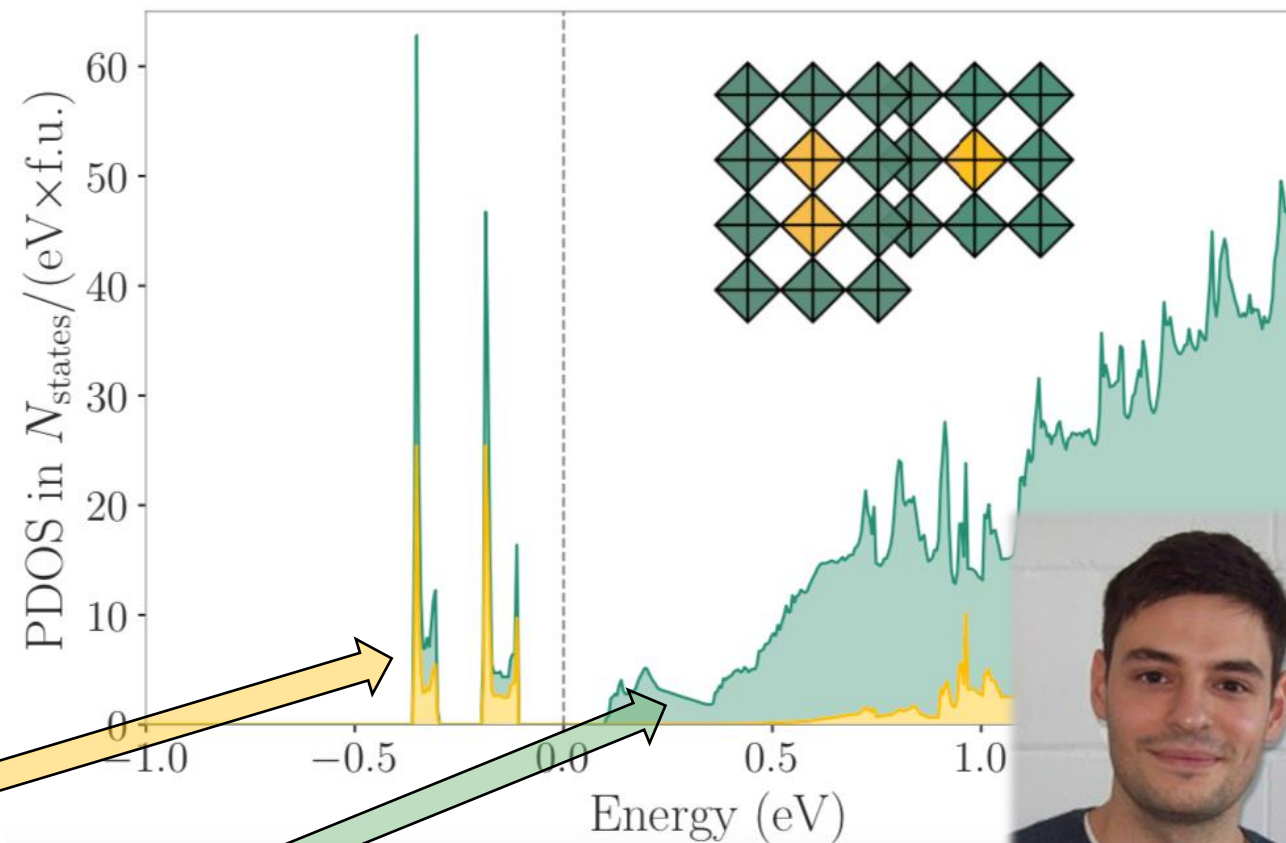


Insights from electronic structure calculations

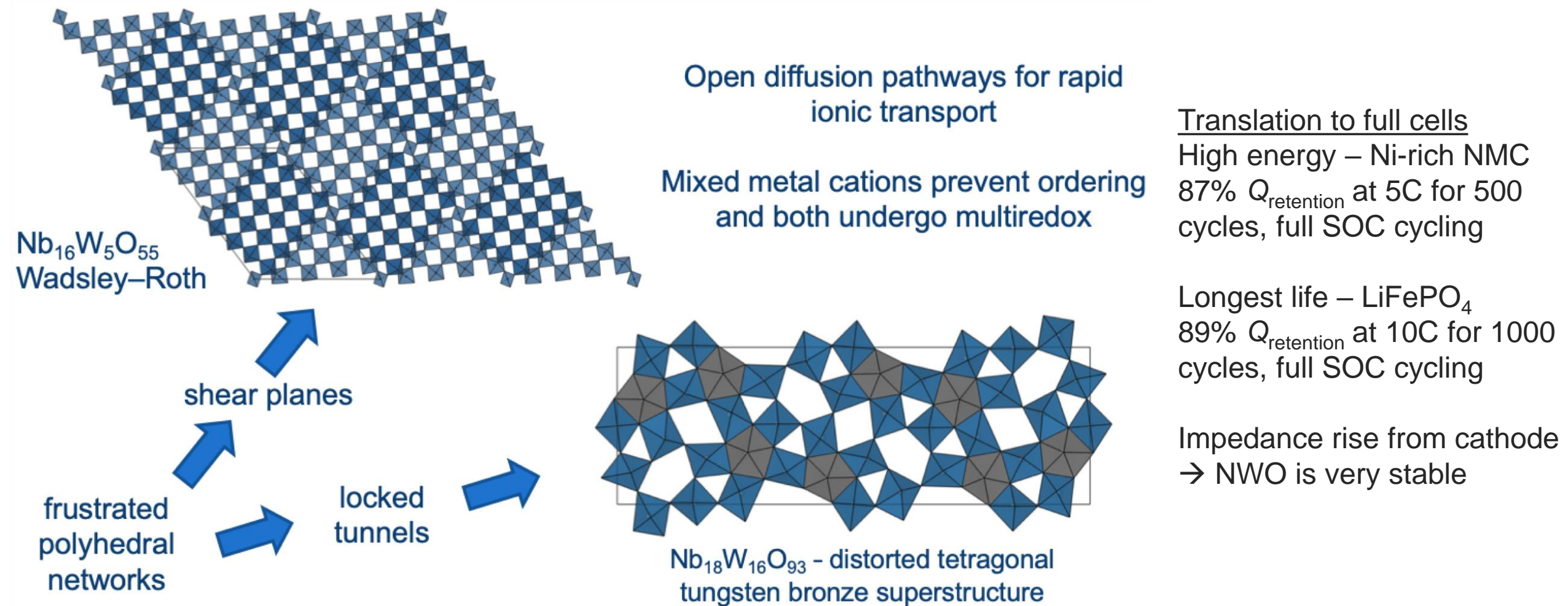


Center of blocks → localized electrons

Crystallographic shear planes → conduction electrons



Mechanism of high-rate Li intercalation in niobium tungsten oxides



Acknowledgements

Clare Grey

Lauren Marbella

Kamila Wiaderek, Giannantonio Cibin, Anatoliy Senyshyn

John Griffin, Alex Forse

Can Koçer, Martin Mayo, Matthew Evans, Chris Pickard, Andrew Morris



*Herchel Smith
Scholarship*

