



Niobium tungsten oxides for high-rate lithium-ion energy storage

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Nature 2018, 559, 556–563.

Electrochemical energy storage

UK set to ban petrol and diesel vehicle sales from 2040

£65 million Faraday Institution for advanced batteries

Grid-scale renewables are increasing and require storage/shifting

Personal electronics, power tools, internet-of-things (IoT), robotics

Lithium-ion battery market (cell level) $\begin{array}{c}
\end{array}
2018 \rightarrow \$31 \text{ billion, 160 GWh} \\
\end{array}
2025 \rightarrow \$80 \text{ billion, 600 GWh} \\
\end{array}
2030 \rightarrow \$140 \text{ billion, 1200 GWh}$





Battery Applications





Images: Toshiba, Chevy Bolt EV, Wall Street Journal, Stanley Black and Decker

Lithium-ion batteries





Pecher, O.; González, J. C.; Griffith, K. J.; Grey, C. P. Materials' Methods: NMR in Battery Research. *Chem. Mater.* **2017**, *29*, 213–242.

State-of-the-art in high power anodes

Lithium titanate spinel: $Li_4Ti_5O_{12}$, LTO

Voltage vs. Li⁺/Li: 1.55 V \rightarrow safety, lower energy Max. theoretical capacity (3 Li/5 Ti): 175 mA·h·g⁻¹ (less in practice) Long cycle life: >15,000 cycles Limited Li⁺ diffusion & e⁻ conductivity \rightarrow nanoscale

Commercial: small anode market share but 25% CAGR 4200 tons/y (2018) → 50,000 tons/y (2030)



Improved high-rate anodes are desired for safe, long lasting, fast charging batteries $TiNb_2O_7$ (Toshiba), crystallographic shear structure

CAGR = compound annual growth rate

Market data: C. Pillot, Avicenne Energy

Niobium-based mixed metal oxides from lessons learnt on Nb₂O₅





Griffith, Kent. J.; Forse, A. C.; Griffin, J. M.; Grey, C. P. High-Rate Intercalation without Nanostructuring in Metastable Nb₂O₅ Bronze Phases. *J. Am. Chem. Soc.* **2016**, *138*, 8888-8899.

Nb₁₆W₅O₅₅ crystal structure





Griffith, K. J.; Wiaderek, K. M.; Cibin, G.; Marbella, L. E.; Grey C. P. Niobium Tungsten Oxides for High-rate Lithium-ion Energy Storage. *Nature*, **2018**, *559*, 556–563.

New anode materials for high power, fast charging lithium-ion batteries

Niobium-based mixed metal oxides from lessons learnt on Nb₂O₅





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Nb₁₈W₁₆O₉₃ crystal structure





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Micrometer-scale bulk particle morphology (for high rates??)



Material synthesis Scalable Low manufacturing cost (Li-free synthesis) Electrode manufacturing Standard powder mixing Standard slurry coating





Battery performance Low surface area = low reactivity → long cycle life, high safety



Niobium tungsten oxide electrochemistry





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0.8

0.6

0.4

Potentia

50

Niobium tungsten oxide electrochemistry

CH H



dashed lines = theoretical one electron per transition metal capacity

Niobium tungsten oxide electrochemistry







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Chemical and structural insights from synchrotron X-rays





Diamond Light Source, Beamline B18 Principal beamline scientist: Giannantonio Cibin

Multi-edge X-ray absorption spectroscopy



XAS: Element specific, sensitive to bulk, electronic and atomic probe



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$Nb_{16}W_5O_{55}XAS @ Nb K, W L_{II}, W L_{I}edges$





Multi-electron Redox at Nb and W





Operando high-rate structure evolution from synchrotron diffraction





Advanced Photon Source, Argonne National Lab; Beamline scientist: Kamila Wiaderek Borkiewicz, O. J.; Shyam, B.; Wiaderek, K. M.; *et al. J. Appl. Cryst.* **2012**, *45*, 1261–1269.

Operando high-rate structure evolution from synchrotron diffraction





Pulsed field gradient NMR Spectroscopy



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<u>Niobium tungsten oxides</u> D_{Li} (298 K) ~ 10⁻¹³ m²·s⁻¹; E_a ~ 0.2–0.3 eV

Putting diffusion coefficients into context

	Diffusion Length (µm)							
D _{Li} (m ² ·s ⁻¹)	1C (3600 s)	20C (180 s)	60C (60 s)					
1.0×10 ⁻¹²	150	33	19					
1.0×10 ⁻¹⁴	15	3.3	1.9					
1.0×10 ⁻¹⁶	1.5	0.33	0.19					
1.0×10 ⁻¹⁸	0.15	0.033	0.019					
1.0×10 ⁻²⁰	0.015	0.0033	0.0019					



on Length	(μm)	Compound	Structure Type	D _{Li} (m ² ·s ⁻¹)	т (к)	Tech- nique	Reference
2 0C (180 s) 33	60C (60 s) 19	Li ₁₀ GeP ₂ S _{12,} Li ₇ GePS _{8,} Li ₁₀ SnP ₂ S ₁₂ Li ₇ P ₃ S ₁₁ , & Li ₁₁ Si ₂ PS ₁₂	Thio-LISICON	1–5 ×10 ⁻¹²	298	PFG NMR	Kuhn et al. (2013), (2014), Hayamizu et al. (2013)
3.3).33	1.9 0.19	β-Li ₃ PS ₄	Thio-LISICON	5.4×10 ⁻¹³	373	PFG NMR	Gobet et al.
).033	0.019	Li _{0.6} [Li _{0.2} Sn _{0.8} S ₂]	Layered (O1)	2-20×10 ⁻¹²	298	PFG NMR	Holzmann et al.
).0033	0.0019	Li _{1.5} Al _{0.5} Ge _{1.5} (PO ₄) ₃	NASICON	2.9×10 ⁻¹³	311	PFG NMR	Hayamizu et al.
		Li _{6.6} La ₃ Zr _{1.6} Ta _{0.4} O ₁₂	Garnet	3.5×10 ⁻¹³	353	PFG NMR	Hayamizu et al.
Liquid electrolytes are 10 ^{−10} −10 ^{−12} m ² ·s ^{−1}		Graphite (Stage I)	Graphite	1-2×10 ⁻¹⁵	298	NMR relaxn.	Langer et al.
		Li ₄ Ti ₅ O ₁₂	Spinel	3.2×10 ⁻¹⁵	298	µ⁺-SR	Sugiyama et al.
		LiMn ₂ O ₄	Spinel	1×10 ⁻²⁰	350	NMR relaxn.	Verhoevenm et al.



Niobium tungsten oxides D_{Li} (298 K) ~ 10⁻¹³ m²·s⁻¹; E_a ~ 0.2–0.3 eV

Insights from electronic structure calculations





 Koçer, Can P.; Griffith, Kent J.; Grey, Clare P.; Morris, Andrew J. *Phys. Rev. B* 2019, *99*, 075151.
 Koçer, Can P.; Griffith, Kent J.; Grey, Clare P.; Morris, Andrew J. Cation Disorder and Lithium Insertion Mechanism of Wadsley–Roth Crystallographic Shear Phases from First Principles. arXiv: 1906.04192

Mechanism of high-rate Li intercalation in niobium tungsten oxides





Griffith, K. J.; Wiaderek, K. M.; Cibin, G.; Marbella, L. E.; Grey C. P. *Nature*, **2018**, *559*, 556–563.
 Kim, Yumi; Griffith, Kent J.; Lee, Jeongjae; Jacquet, Quentin; Rinkel, Bernardine L. D.; Grey, Clare P. High Rate Lithium Ion Battery with Niobium Tungsten Oxide Anode. *In preparation.*

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